



Name: _____
Date: _____

COUNTING ATOMS

WORKSHEET

- Chemical formulas can have three components.
 - The _____ which are represented by a single uppercase letter, or represented by an uppercase letter followed by a lowercase letter. This tells you the types of elements in the compound.
 - The _____ which are numbers that are found on the lower _____ - hand side of each element symbol. This tells you the number of _____ of this element in the molecule. If an element symbol has no subscript next to it, then this indicates that there is only _____ atom of this element in the molecule.
 - The _____ which surround some groups of atoms indicate that the numbers of all of the atoms inside the bracket need to be _____ by the _____ on the outside of the bracket found on the lower _____ - hand side of that bracket.
- Record the number of each atom in each molecule, then record the total number of atoms in the molecule:

1) NaOH	
Element	# of Atoms
Total	

4) Li ₂ SO ₄	
Element	# of Atoms
Total	

7) H ₂ COCH ₃	
Element	# of Atoms
Total	

10) NH ₄ C ₂ H ₃ O ₂	
Element	# of Atoms
Total	

13) Ca ₃ (PO ₄) ₂	
Element	# of Atoms
Total	

2) HNO ₃	
Element	# of Atoms
Total	

5) NH ₄ Cl	
Element	# of Atoms
Total	

8) Mg(OH) ₂	
Element	# of Atoms
Total	

11) NaC ₂ H ₃ O ₂	
Element	# of Atoms
Total	

14) Fe ₂ (CO ₃) ₃	
Element	# of Atoms
Total	

3) U ₂ O	
Element	# of Atoms
Total	

6) CaClO ₄	
Element	# of Atoms
Total	

9) Al(OH) ₃	
Element	# of Atoms
Total	

12) (NH ₄) ₃ PO ₄	
Element	# of Atoms
Total	

15) Al ₂ (SO ₄) ₃	
Element	# of Atoms
Total	

Answer Key For Counting Atoms In Compounds

**Tina Overton, Jonathan Rourke, Fraser
A. Armstrong**



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