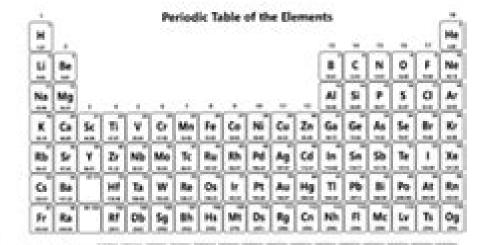
Name: Period:

# Select the correct answers by circling one from each letter.

- 1. Which of the following in each pair has the smallest radius?
  - a. P O (circle one)
  - b. Te I (circle one)
  - c. Ne Kr (circle one)
  - d. Na S (circle one)
- 2. Which of the following in each group has the largest ionization energy?
  - a. S Cl Ar (circle one)
  - b. K Ca Rb (circle one)
  - c. Ar Xe Kr (circle one)
  - d. O N C (circle one)
- 3. Which has the highest electronegativity in each group?
  - a. B C N (circle one)
  - b. B Al Ge (circle one)
  - e. Ge O P (circle one)
  - d. Se S Cl (circle one)
- 4. Name three metal families on the periodic table.
  - 3.
  - Ъ.



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**G Psacharopoulos** 

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