

Academic

Periodic Trends Worksheet 2

Names:

Period:

Select the correct answers by circling one from each letter.

1. Which of the following in each pair has the smallest radius?

- a. P O (circle one)
b. Te I (circle one)
c. Ne Kr (circle one)
d. Na S (circle one)

2. Which of the following in each group has the largest ionization energy?

- a. S Cl Ar (circle one)
b. K Ca Rb (circle one)
c. Ar Xe Kr (circle one)
d. O N C (circle one)

3. Which has the highest electronegativity in each group?

- a. B C N (circle one)
b. B Al Ge (circle one)
c. Ge O P (circle one)
d. Se S Cl (circle one)

4. Name three metal families on the periodic table.

-

1 H																	18 Ne				
3 Li	4 Be																	7 N	8 O	9 F	10 Ne
11 Na	12 Mg																	15 P	16 S	17 Cl	18 Ar
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr				
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe				
55 Cs	56 Ba		57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu				
87 Fr	88 Ra		89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr				

Answer Key Quick Lab Periodic Table In Ionic Radii

G Psacharopoulos



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