

Chemistry – Unit 6 Worksheet 1

We have observed evidence that atoms of different elements are different, the most noticeable being their predictable characteristic chemical behaviour. Atoms of one element react in a predictable way with atoms of another element, whereas atoms of another element react in a predictable way with atoms of a third element, but not with atoms of the first element. We will now examine the patterns that reveal the way atoms combine when they form a compound, as well as the way they change at the same time.

1. Write the formula and draw the particle diagram for each compound. The ratio of atoms in each compound is given.

Atomic formula Compound	12 hydrogen 16 oxygen H_2O	12 hydrogen 16 oxygen H_2O_2	12 hydrogen 16 oxygen H_2O_2	1 hydrogen 1 oxygen H_2O
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particle



Diagram

Atomic formula Compound	12 hydrogen 16 oxygen H_2O	12 hydrogen 16 oxygen H_2O_2	12 hydrogen 16 oxygen H_2O_2	1 hydrogen 1 oxygen H_2O
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particle



Diagram

Atomic formula Compound	12 hydrogen 16 oxygen H_2O	12 hydrogen 16 oxygen H_2O_2	1 hydrogen 16 oxygen H_2O	1 hydrogen 16 oxygen H_2O
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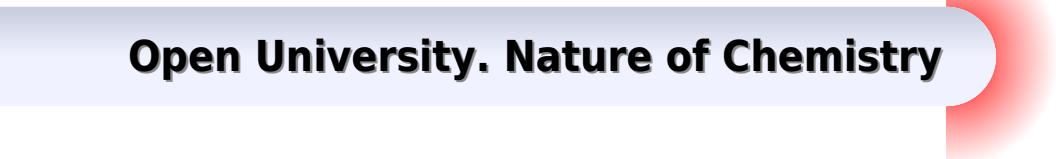
particle



Diagram

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