
ACS General Chemistry practice 147 Questions with Verified Answers

Examples of Lewis-Acid based reactions - CORRECT ANSWER $\text{H}^+(\text{aq}) + \text{OH}^-(\text{aq}) \rightarrow \leftarrow \text{H}_2\text{O}(\text{l})$,
 $\text{HCl}(\text{g}) + \text{NH}_3(\text{g}) \rightarrow \text{NH}_4\text{Cl}(\text{s})$
 $\text{H}^+(\text{aq}) + \text{OH}^-(\text{aq}) \rightarrow \leftarrow \text{H}_2\text{O}(\text{l})$

According to the Lewis definition, an acid is a species - CORRECT ANSWER accepting a pair of electrons.

All are potential Lewis bases except - CORRECT ANSWER CH_4

Which hydrated metal ion is most acidic under conditions of equal molar concentration in water? - CORRECT ANSWER Al^{3+}

Which element in Group 5A forms the most basic oxide? - CORRECT ANSWER Bi

6. In what respect does an atom of magnesium differ from a magnesium ion (Mg^{2+})? - CORRECT ANSWER The ion has a more stable electronic arrangement than the atom.

A certain atom has the symbol $^{148}_{73}\text{P}$.

What does this symbol tell about an atom of the element? - CORRECT ANSWER It has 75 neutrons in the nucleus

The atomic mass of an element is 32.07 u and its atomic number is 16. The number of protons contained in the nucleus of an atom of this element is - CORRECT ANSWER 16

The atomic mass of an element is 32.07 u and its atomic number is 16. The element forms a simple ion. The ion will most likely have a charge of - CORRECT ANSWER $2-$

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Interdisciplinary Approaches to Distance Teaching Alan Blackstock, Nathan Straight, 2015-11-19 Synchronous technologies particularly interactive video conferencing IVC are becoming common modes of teaching and delivering college courses The increasing popularity of IVC in the U S and abroad calls for more pedagogically effective practices for instructors using this technology This volume focuses on innovative and proven approaches to IVC teaching in a variety of disciplines English history biology chemistry geology engineering social work and elementary and special education Contributors hail from a pioneering university at the forefront of distance education and understand the practice and potential of IVC teaching at the highest levels Chapters outline the challenges and benefits of IVC teaching from pedagogical technical and administrative perspectives

Why We Teach Science John L. Rudolph, 2022-12-20 Few people question the importance of science education in American schooling The public readily accepts that it is the key to economic growth through innovation develops the ability to reason more effectively and enables us to solve the everyday problems we encounter through knowing how the world works Good science teaching results in all these benefits and more or so we think But what if all this is simply wrong What if the benefits we assume science education produces turn out to be an illusion nothing more than wishful thinking In Why We Teach Science and Why We Should former high school teacher and historian of science education John L Rudolph examines the reasons we've long given for teaching science and assesses how they hold up to what we know about what students really learn or don't learn in science classrooms and what research tells us about how people actually interact with science in their daily lives The results will surprise you Instead of more and more rigorous traditional science education to fill the STEM pipeline Rudolph challenges us to think outside the box and makes the case for an expansive science education aimed instead at rebuilding trust between science and the public something we desperately need in our current era of impending natural challenges and science denial

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Preparing for Your ACS Examination in General Chemistry - the Official Guide LUCY T. EUBANKS, 2018-10-15 This guide is separated into first term and second term general chemistry material Each section contains 8 chapters of material that also aligns to most general chemistry textbooks for a seamless addition to study materials for students Each chapter is designed with an introductory section of the material including common representations and where to find this material in a textbook The second section provides worked examples of typical multiple choice questions including how the correct answer is determined as well as how the incorrect answers were determined Also included for each study problem is a listing of the corresponding practice questions that use that concept The final section is a series of practice problems to test the concepts collectively The key is provided on a separate page for all study and practice problems

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