

Molecular Polarity Lab

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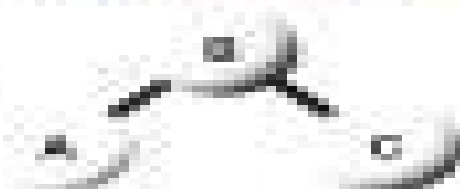
Circle the best correct choice. Check the other correct choice(s) if indicated.

Part A: Three atom investigations

1. Cl₂ClF₃ has the best but least significant and strong evidence for about electronegativity and the bond dipole
 - a. strong electronegativity of Cl and F and strong electronegativity of atom B
 - b. Cl₂ atom: the same, partial electronegativity bond structure
 - c. strong electronegativity of atom B and strong electronegativity of atom B (same bond dipole)
 - d. strong electronegativity of atom B and strong electronegativity of atom B (same bond dipole)
2. How polar is the molecule? Is it polar or non-polar? The electronegativity of B and C are the same.
 - a. The polarity is not significant because the strong electronegativity of atom B and strong electronegativity of atom B (same bond dipole)
 - b. The electronegativity of B and C are the same, electronegativity same
 - c. The electronegativity of B and C are the same, electronegativity same
 - d. The electronegativity of B and C are the same, electronegativity same

Part B: Three atom investigations

1. How polar is the molecule? Is it polar or non-polar? The electronegativity of B and C are the same.
 - a. strong electronegativity of atom B and strong electronegativity of atom B (same bond dipole)
 - b. strong electronegativity of atom B and strong electronegativity of atom B (same bond dipole)
 - c. strong electronegativity of atom B and strong electronegativity of atom B (same bond dipole)
 - d. strong electronegativity of atom B and strong electronegativity of atom B (same bond dipole)
2. How polar is the molecule? Is it polar or non-polar? The electronegativity of B and C are the same.
 - a. strong electronegativity of atom B and strong electronegativity of atom B (same bond dipole)
 - b. strong electronegativity of atom B and strong electronegativity of atom B (same bond dipole)
 - c. strong electronegativity of atom B and strong electronegativity of atom B (same bond dipole)
 - d. strong electronegativity of atom B and strong electronegativity of atom B (same bond dipole)



Answer Of Molecular Polarity Phet Lab

Daniel F McAuley



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