

MOLAR SPECIFIC HEAT OF GASES:

1. $\Delta U = C_V \Delta T$
2. $C_p - C_v = R$
3. $C_v = R/\gamma - 1$
4. $C_p = \gamma R/\gamma - 1$
5. $C_v = 1/2 f R$ where f = degree of freedom
6. $C_p = (f + 2/2)R$
7. $\gamma = C_p/C_v$

THERMODYNAMIC SCALE:

1. $T = 273.16 Q_3/Q_2$
2. $Q_2/Q_3 = T_2/273.16 K$
3. $\frac{Q_1 - Q_2}{Q_2} = \frac{P_1 - P_2}{P_2} = \frac{V_1 - V_2}{V_2}$
4. $\frac{\Delta T}{100} = \frac{\Delta T}{100} = \frac{\Delta T}{100}$

2nd year MDCAT FORMULA LIST PHYSICS

ELECTROSTATICS

COULOMB'S LAW:

1. $F = k \frac{Q_1 Q_2}{r^2}$
2. $k = \frac{1}{4\pi\epsilon_0}$
3. $F = \frac{1}{4\pi\epsilon_0} \frac{Q_1 Q_2}{r^2}$
4. $E = \frac{F}{q}$
5. $\epsilon_0 = \epsilon_r \epsilon_0$
6. $F = \frac{1}{4\pi\epsilon_0} \frac{Q_1 Q_2}{r^2}$
7. $F = \frac{Q_1 Q_2}{4\pi\epsilon_0 r^2}$

ELECTRIC FIELD INTENSITY:

1. $E = \frac{F}{q}$
2. $E = \frac{1}{4\pi\epsilon_0} \frac{Q}{r^2}$
3. $E = \frac{1}{4\pi\epsilon_0} \frac{Q}{r^2}$
4. $E = \frac{1}{4\pi\epsilon_0} \frac{Q}{r^2}$
5. $E = \frac{Q}{4\pi\epsilon_0 r^2}$
6. $E = \frac{Q}{4\pi\epsilon_0 r^2}$
7. $E = \frac{Q}{4\pi\epsilon_0 r^2} = E = \frac{Q}{4\pi\epsilon_0 r^2}$

110011 Pysics Priveous Year Paper

J. Bee Bednar



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Table of Contents 110011 Pysics Priveous Year Paper

1. Understanding the eBook 110011 Pysics Priveous Year Paper
 - The Rise of Digital Reading 110011 Pysics Priveous Year Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying 110011 Pysics Priveous Year Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 110011 Pysics Priveous Year Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from 110011 Pysics Priveous Year Paper
 - Personalized Recommendations
 - 110011 Pysics Priveous Year Paper User Reviews and Ratings
 - 110011 Pysics Priveous Year Paper and Bestseller Lists

5. Accessing 110011 Pysics Priveous Year Paper Free and Paid eBooks
 - 110011 Pysics Priveous Year Paper Public Domain eBooks
 - 110011 Pysics Priveous Year Paper eBook Subscription Services
 - 110011 Pysics Priveous Year Paper Budget-Friendly Options
6. Navigating 110011 Pysics Priveous Year Paper eBook Formats
 - ePub, PDF, MOBI, and More
 - 110011 Pysics Priveous Year Paper Compatibility with Devices
 - 110011 Pysics Priveous Year Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 110011 Pysics Priveous Year Paper
 - Highlighting and Note-Taking 110011 Pysics Priveous Year Paper
 - Interactive Elements 110011 Pysics Priveous Year Paper
8. Staying Engaged with 110011 Pysics Priveous Year Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 110011 Pysics Priveous Year Paper
9. Balancing eBooks and Physical Books 110011 Pysics Priveous Year Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 110011 Pysics Priveous Year Paper
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 110011 Pysics Priveous Year Paper
 - Setting Reading Goals 110011 Pysics Priveous Year Paper
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 110011 Pysics Priveous Year Paper
 - Fact-Checking eBook Content of 110011 Pysics Priveous Year Paper
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

-
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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