Write a balanced chemical equation to represent each of the following chemical reactions.

- 1. iron + sulfur iron(II) sulfide
- 2. zinc + copper(II) sulfate → zinc sulfate + copper
- 3. silver nitrate + sodium bromide → sodium nitrate + silver bromide
- potassium chlorate (heated) potassium chloride + oxygen
- water (electricity) → hydrogen + oxygen
- mercury(II) oxide (heated) → mercury + oxygen
- potassium iodide + lead(II) nitrate → lead(II) iodide + potassium nitrate
- 8. aluminum + oxygen → aluminum oxide
- 9. magnesium chloride + ammonium nitrate -- magnesium nitrate + ammonium chloride
- 10. iron(III) chloride + ammonium hydroxide → iron(III) hydroxide + ammonium chloride
- sodium peroxide + water → sodium hydroxide + oxygen
- iron(III) oxide + carbon → iron + carbon monoxide
- 13. iron + water → hydrogen + iron(III) oxide
- iron(III) chloride + potassium hydroxide → potassium chloride + iron(III) hydroxide
- 15. aluminum + sulfuric acid → aluminum sulfate + hydrogen
- 16. sodium carbonate + calcium hydroxide → sodium hydroxide + calcium carbonate

Answer Key Weston Walch Chemistry

Camilla Rothe

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