

Electrostatic Potential Lab

Name _____ Per. _____

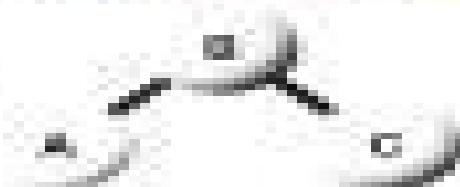
Circle the molecule below: Circle the ring carbon in the structure.

Part 1: Three atom investigation

1. Click the **Keep the box** but **don't display structure** and **show** molecules for about three molecules and the bond charges.
2. Change the electronegativity of **A** and **change** electronegativity of **carbon B**. Observe the arrow, partial charges and bond structure. Write the following information: the electronegativity of atom B (use last digit)
 - atom _____
 - partial charges _____
 - bond structure _____
3. Show partial charge polarity for a carbon nitrate containing electronegativity of B and observe the results. Fill in the following information:
 - a. The polarity arrow points to the _____ electronegativity.
 - b. The partial positive charge is always on the _____ electronegative atom.
 - c. The larger the electronegativity difference more _____ the bond structure.

Part 2: Three atom investigation

1. Change the central atom electronegativity and the electronegativity of **A** and **carbon B**. Circle to show the bond dipole, molecular dipole and partial charges.
 - a. Draw the bond dipole on the Lewis dot diagram.
 - b. In a different color, describe partial molecular dipole on the diagram.
 - c. Draw the partial charge symbolical position on the diagram.
 - d. Circle the electronegative. Add other electronegative the molecule, for example.
2. Clicking more atoms.  Change the **Keep** electronegativity and the electronegativity of **A** and **carbon B** and **carbon C**. Circle to show the bond dipole, molecular dipole and partial charges.
 - a. Draw the bond dipole on the Lewis dot diagram.
 - b. In a different color describe partial molecular dipole on the diagram.
 - c. Draw the partial charge symbolical position on the diagram.
 - d. Circle the electronegative. Describe other electronegative the molecule, for example.



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Table of Contents Answer Of Molecular Polarity Phet Lab

1. Understanding the eBook Answer Of Molecular Polarity Phet Lab
 - The Rise of Digital Reading Answer Of Molecular Polarity Phet Lab
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Of Molecular Polarity Phet Lab
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Of Molecular Polarity Phet Lab
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Of Molecular Polarity Phet Lab
 - Personalized Recommendations
 - Answer Of Molecular Polarity Phet Lab User Reviews and Ratings
 - Answer Of Molecular Polarity Phet Lab and Bestseller Lists

5. Accessing Answer Of Molecular Polarity Phet Lab Free and Paid eBooks
 - Answer Of Molecular Polarity Phet Lab Public Domain eBooks
 - Answer Of Molecular Polarity Phet Lab eBook Subscription Services
 - Answer Of Molecular Polarity Phet Lab Budget-Friendly Options
6. Navigating Answer Of Molecular Polarity Phet Lab eBook Formats
 - ePub, PDF, MOBI, and More
 - Answer Of Molecular Polarity Phet Lab Compatibility with Devices
 - Answer Of Molecular Polarity Phet Lab Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer Of Molecular Polarity Phet Lab
 - Highlighting and Note-Taking Answer Of Molecular Polarity Phet Lab
 - Interactive Elements Answer Of Molecular Polarity Phet Lab
8. Staying Engaged with Answer Of Molecular Polarity Phet Lab
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer Of Molecular Polarity Phet Lab
9. Balancing eBooks and Physical Books Answer Of Molecular Polarity Phet Lab
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Of Molecular Polarity Phet Lab
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answer Of Molecular Polarity Phet Lab
 - Setting Reading Goals Answer Of Molecular Polarity Phet Lab
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answer Of Molecular Polarity Phet Lab
 - Fact-Checking eBook Content of Answer Of Molecular Polarity Phet Lab
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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