

Ternary compounds have _____ (three) _____ (three or) elements. They are usually always composed of a monatomic _____ (monatomic) anionic ion and a polyatomic _____ (polyatomic) anion. The only positively charged polyatomic ion is _____ (ammonium). To write the formula of a ternary compound is no different than to write the formula of a binary compound with one exception. If a subscript is necessary for the polyatomic ion in order to _____ (balance) charge, we must place the polyatomic ion in _____ (parentheses).

As each ion, write the formula of the ionic compound consisting of the anion-ion on the left of the formula and the cation-ion on the right.

	OH^-	SO_4^{2-}	NO_3^-	CO_3^{2-}	PO_4^{3-}
K^+	_____	_____	_____	_____	_____
Na^{2+}	_____	_____	_____	_____	_____
NH_4^+	_____	_____	_____	_____	_____
Al^{3+}	_____	_____	_____	_____	_____
Ca^{2+}	_____	_____	_____	_____	_____
Fe^{2+}	_____	_____	_____	_____	_____

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Table of Contents Answers For Writing Ternary Formulas

1. Understanding the eBook Answers For Writing Ternary Formulas
 - The Rise of Digital Reading Answers For Writing Ternary Formulas
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers For Writing Ternary Formulas
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers For Writing Ternary Formulas
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers For Writing Ternary Formulas
 - Personalized Recommendations
 - Answers For Writing Ternary Formulas User Reviews and Ratings

- Answers For Writing Ternary Formulas and Bestseller Lists
- 5. Accessing Answers For Writing Ternary Formulas Free and Paid eBooks
 - Answers For Writing Ternary Formulas Public Domain eBooks
 - Answers For Writing Ternary Formulas eBook Subscription Services
 - Answers For Writing Ternary Formulas Budget-Friendly Options
- 6. Navigating Answers For Writing Ternary Formulas eBook Formats
 - ePub, PDF, MOBI, and More
 - Answers For Writing Ternary Formulas Compatibility with Devices
 - Answers For Writing Ternary Formulas Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers For Writing Ternary Formulas
 - Highlighting and Note-Taking Answers For Writing Ternary Formulas
 - Interactive Elements Answers For Writing Ternary Formulas
- 8. Staying Engaged with Answers For Writing Ternary Formulas
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers For Writing Ternary Formulas
- 9. Balancing eBooks and Physical Books Answers For Writing Ternary Formulas
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers For Writing Ternary Formulas
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answers For Writing Ternary Formulas
 - Setting Reading Goals Answers For Writing Ternary Formulas
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answers For Writing Ternary Formulas
 - Fact-Checking eBook Content of Answers For Writing Ternary Formulas
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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