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Date: \_\_\_\_\_

## Student Exploration: Covalent Bonds

**Vocabulary:** covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

1. There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers? Share

2. Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?

Share 2

### Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the *Covalent Bonds* Gizmo, atoms form bonds in this way.

To begin, check that **Fluorine** is selected from the **Select a substance** menu. Click **Play** () to see the electrons orbiting the nucleus of each atom.



1. The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have? 7

2. Click **Pause** (). Drag an electron from the left atom to the right atom. Click **Play**.

What happens? it orbits both atoms

3. Click **Pause**, drag an electron from the right atom to the left, and then click **Play**.

What happens now? it orbits both atoms

# Answers For Student Exploration Covalent Bonds

**CH Cherryholmes**



## **Answers For Student Exploration Covalent Bonds:**

**Exploration of the Structure of Atom** Chandan Sengupta, First Publication April 2019 Revised Publication October 2022 Second Revised Edition July 2023 Third Revised Edition December 2024 Published by Chandan Sengupta Printed by IECIT Printing and Publication Services Resource Centre Arabinda Nagar Bankura 722101 W B India Topics Covered Atoms and Molecules Structure of Atom Bonding Mechanism and Chemical Reactions Mechanism of Bonding This book is suitable for students of Class 9 to Class 11 Students aspiring for Pre Medical Entrance Examination can also get adequate support Additional Hard Copies can also be obtained from Chandan Sukumar Sengupta Arabinda Nagar Bankura 722101 WB Write to Us for more materials

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Barbara Houtz,2011-06-01 Developed for grades 6-12 this rich resource provides teachers with practical strategies to enhance science instruction Strategies and model lessons are provided in each of the following overarching topics inquiry and exploration critical thinking and questioning real world applications integrating the content areas and technology and assessment Research based information and management techniques are also provided to support teachers as they implement the strategies within this resource This resource supports core concepts of STEM instruction **AI in**

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*Teaching the Content Areas to English Language Learners in Secondary Schools* Luciana C. de Oliveira,Kathryn M. Obenchain,Rachael H. Kenney,Alandeom W. Oliveira,2019-01-17 This practitioner based book provides different approaches for reaching an increasing population in today s schools English language learners ELLs The recent development and adoption of the Common Core State Standards for English Language Arts and Literacy in History Social Studies Science and Technical Subjects CCSS ELA Literacy the Common Core State Standards for Mathematics the C3 Framework and the Next Generation Science Standards NGSS highlight the role that teachers have in developing discipline specific competencies This requires new and innovative approaches for teaching the content areas to all students The book begins with an introduction that contextualizes the chapters in which the editors highlight transdisciplinary theories and approaches that cut across content areas In addition the editors include a table that provides a matrix of how strategies and theories map across the chapters The four sections of the book represent the following contentareas English language arts mathematics science and social studies This book offers practical guidance that is grounded in relevant theory and research and offers teachers suggestions on how to use the approaches described

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**Chemical Interactions**, 2005 **Never Lose Your Nerve!** Alan J. Heeger, 2015-10-09 Never Lose Your Nerve chronicles the ups and downs of a Nobel Laureate s life Professor Alan J Heeger was awarded the Nobel Prize in Chemistry in 2000 together with Professor Alan G MacDiarmid and Professor Hideki Shirakawa Filled with humor this book tells Professor Heeger s story his love for his family especially how his wife s love has always been his guiding light his progress from a young student to an eminent scientist his passion for the theatre and its impact on his science his adventures as a successful entrepreneur and his personal losses Many think of scientists as risk adverse individuals but Professor Heeger shows the absolute necessity of risk in research and that scientists are in fact risk addicted as taking the first risky step into unfamiliar territory is a step in the right direction towards creativity and great discoveries Never lose your nerve and you will be rewarded Life is an exciting adventure and this book clearly demonstrates it and is for those who are looking to impact others Perhaps the greatest pleasure of being a scientist is to have an abstract idea then to do an experiment more often a series of experiments is required that demonstrates the idea was correct that is Nature actually behaves as conceived in the mind of the scientist This process is the essence of creativity in science I have been fortunate to have experienced this intense pleasure many times in my life Alan J Heeger

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