

Experiment 10: Electrodes with a Standard Solution

From: *Basic Laboratory Experiments*, Experiment 10, 45-50

The purpose of this experiment is to determine the standard potential for several cells consisting of $\text{Zn} / \text{Zn}^{2+}$ and $\text{Cu} / \text{Cu}^{2+}$ electrodes. Electrode potentials are a half cell concept that allows us to compare standard reduction potentials relative to a common reference electrode. In this experiment, we will use a standard hydrogen electrode (SHE) as a reference electrode. Other standard electrodes, such as the silver-silver chloride electrode and the saturated calomel electrode, are often used in place of the SHE. The standard potential for the SHE is defined as 0.00 V.

In this experiment, you will determine the standard potential for the $\text{Zn} / \text{Zn}^{2+}$ and $\text{Cu} / \text{Cu}^{2+}$ electrodes. You will also determine the standard potential for the $\text{Zn} / \text{Zn}^{2+}$ and $\text{Cu} / \text{Cu}^{2+}$ electrodes.

Objectives:

- to determine the standard potential for the $\text{Zn} / \text{Zn}^{2+}$ and $\text{Cu} / \text{Cu}^{2+}$ electrodes
- to determine the standard potential for the $\text{Zn} / \text{Zn}^{2+}$ and $\text{Cu} / \text{Cu}^{2+}$ electrodes
- to determine the standard potential for the $\text{Zn} / \text{Zn}^{2+}$ and $\text{Cu} / \text{Cu}^{2+}$ electrodes
- to determine the standard potential for the $\text{Zn} / \text{Zn}^{2+}$ and $\text{Cu} / \text{Cu}^{2+}$ electrodes
- to determine the standard potential for the $\text{Zn} / \text{Zn}^{2+}$ and $\text{Cu} / \text{Cu}^{2+}$ electrodes

Materials:

- 1.0 M ZnSO_4 solution
- 1.0 M CuSO_4 solution
- 1.0 M H_2SO_4 solution
- 1.0 M KNO_3 solution
- 1.0 M NaOH solution
- 1.0 M HCl solution
- 1.0 M H_2O_2 solution
- 1.0 M H_2O_2 solution
- 1.0 M H_2O_2 solution

Procedure:

- 1.0 M ZnSO_4 solution
- 1.0 M CuSO_4 solution
- 1.0 M H_2SO_4 solution
- 1.0 M KNO_3 solution
- 1.0 M NaOH solution
- 1.0 M HCl solution
- 1.0 M H_2O_2 solution
- 1.0 M H_2O_2 solution
- 1.0 M H_2O_2 solution

Results:

- 1.0 M ZnSO_4 solution
- 1.0 M CuSO_4 solution
- 1.0 M H_2SO_4 solution
- 1.0 M KNO_3 solution
- 1.0 M NaOH solution
- 1.0 M HCl solution
- 1.0 M H_2O_2 solution
- 1.0 M H_2O_2 solution
- 1.0 M H_2O_2 solution

Discussion:

- 1.0 M ZnSO_4 solution
- 1.0 M CuSO_4 solution
- 1.0 M H_2SO_4 solution
- 1.0 M KNO_3 solution
- 1.0 M NaOH solution
- 1.0 M HCl solution
- 1.0 M H_2O_2 solution
- 1.0 M H_2O_2 solution
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Tao Wei



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