HIMACHAL PRADESH BOARD OF SCHOOL EDUCATION, DHARAMSHALA Model Question Paper Class – 11th

Duration - 03:00 Hr

Chemistry(Theory)

M.M.: 60

Instructions:

- The question paper consists of four sections- A, B, C and D.
- ii) Internal choices are given in some questions.
- Section A contains 12 MCQ 1 to 12 of 1 mark each.
- iv) Section B contains 9 questions from 13 to 21 of 2 marks each.
- Section C contains 6 questions from 22 to 27 of 3 marks each.
- vi) Section D contains 3 questions from 28 to 30 of 4 marks each.
- vii) Make neat and clean diagrams where required.

Section A (1X12)

- Q.1 One mole of oxygen molecule is equal to
- a) 6.023x10²³ atoms of oxygen b) 6.023x10⁻²³ atoms of oxygen
- c) 6.023x10²³ molecules of oxygen d) 6.023x10⁻²³ molecules of oxygen
- Q.2 3d orbitals have values
- a) n=2, l=3
- c) n=3, l=1

- b) n=3, l=0
- d) n=3, l=2
- Q. 3 Which of the following elements has maximum electron gain enthalpy
- a) F

b) Cl

- c) Br
- Q.4 Pyramidal geometry associated with
- a) CH₄
 b) NH₃

c) H₂O

d) CO2

d) I

- Q.5 Decreasing order of bond angle in H₂O, NH₃ and CH₄
- a) CH₄>NH₃>H₂O
- b) NH₃>H₂O>CH₄
- c) CH₄>H₂O>NH₃ d) H₂O>CH₄>NH₃
- Q.6 According to Boyle's law
- b) At constant T, V∞ 1/P
- a) At constant T, V∞ P
 c) At constant P, V∞ T
- d) At constant T, P ∞T
- Q.7 For the reaction $N_2(g)+3H_2(g)=2NH_{2(g)}$

Answers For Exercise Of Hsc 11th Chemistry

RS Peters

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