

Name: \_\_\_\_\_

ID#: \_\_\_\_\_

55 multiple-choice questions (3 pts each) + 6 fill-in questions (40 pts) = Total (205 pts).

1) How many **unpaired** electrons are present in the electron configuration of the **O** atom?

- A) none B) one
- C) two**
- D) three E) four

2) Which of the following contain(s) **polar covalent** bonds?

- A)
- $H_2$
- B)
- $NaI$
- C)
- $Cl_2$
- D)  $H_2O$**
- E) both A and C

3) What is the formal charge and hybridization for the highlighted **Carbon**?

- A) 0,
- $sp^2$
- B) +1,
- $sp^2$
- C) -1,  $sp$**
- D) +1,
- $sp$
- E) 0,
- $sp$

4) What of the following best describes the hybridization of C in **carbocation**  $CH_3^+$ ?

- A)
- $sp^3$
- B)  $sp^2$**
- C)
- $sp$
- D) s

5) what orbital is the nitrogen lone pair residing in the following molecule?

- A)
- $sp^3$
- B)  $sp^2$**
- C)
- $sp$
- D) p E) s

6) Identify the **most acidic** carboxylic acid.**Answer is E**7) The **conjugate acid** of  $H_2O$  is \_\_\_\_\_

- A)
- $HO^-$
- B)
- $H_3O$
- C)  $H_3O^+$**
- D)
- $H_2O$
- E)
- $H_2O^+$

8) How many potential stereoisomers exist for the following molecule.

- A) 1 B) 2
- C) 3**
- D) 4 E) 5



9) what is the name of the following compound?

- A) *trans*-4-ethylcyclohexan-1-ol B) *cis*-4-ethylhexan-1-ol  
 C) *trans*-4-methylcyclohexan-1-ol **D) *cis*-4-ethylcyclohexan-1-ol**  
 E) *trans*-4-ethylcyclopentan-1-ol



10) which of the following step is part of the propagation in the methane chlorination?

- A)  $Cl-Cl \longrightarrow 2 Cl \cdot$  B)  $H_3C \cdot + Cl-Cl \longrightarrow H_3C-Cl + Cl \cdot$   
 C)  $H_3C-H + \cdot Cl \longrightarrow H_3C \cdot + H-Cl$  D) B and C

**Answer is D**

11) Based on the Hammond postulate, which of the following transition state best explains the following reaction step?

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1. Understanding the eBook Answer Key Weston Walch Organic Chemistry
  - The Rise of Digital Reading Answer Key Weston Walch Organic Chemistry
  - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key Weston Walch Organic Chemistry
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Answer Key Weston Walch Organic Chemistry
  - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key Weston Walch Organic Chemistry
  - Personalized Recommendations

- Answer Key Weston Walch Organic Chemistry User Reviews and Ratings
- Answer Key Weston Walch Organic Chemistry and Bestseller Lists
- 5. Accessing Answer Key Weston Walch Organic Chemistry Free and Paid eBooks
  - Answer Key Weston Walch Organic Chemistry Public Domain eBooks
  - Answer Key Weston Walch Organic Chemistry eBook Subscription Services
  - Answer Key Weston Walch Organic Chemistry Budget-Friendly Options
- 6. Navigating Answer Key Weston Walch Organic Chemistry eBook Formats
  - ePub, PDF, MOBI, and More
  - Answer Key Weston Walch Organic Chemistry Compatibility with Devices
  - Answer Key Weston Walch Organic Chemistry Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Answer Key Weston Walch Organic Chemistry
  - Highlighting and Note-Taking Answer Key Weston Walch Organic Chemistry
  - Interactive Elements Answer Key Weston Walch Organic Chemistry
- 8. Staying Engaged with Answer Key Weston Walch Organic Chemistry
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Answer Key Weston Walch Organic Chemistry
- 9. Balancing eBooks and Physical Books Answer Key Weston Walch Organic Chemistry
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Answer Key Weston Walch Organic Chemistry
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Answer Key Weston Walch Organic Chemistry
  - Setting Reading Goals Answer Key Weston Walch Organic Chemistry
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answer Key Weston Walch Organic Chemistry
  - Fact-Checking eBook Content of Answer Key Weston Walch Organic Chemistry

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
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