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Journal of the American Medical Association American Medical Association, 1929

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reaction ea is the minimum amount of energy for a reaction to occur it is the amount of energy required to create an activated complex 4 what happens when a catalyst is used in a reaction a catalyst changes the reaction 18 4 potential energy diagrams chemistry libretexts - Oct 04 2023

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web 1 draw a potential energy diagram for an endothermic reaction 2 label your drawing with the following letters a pe of products d energy of activation b pe of reactants e heat of reaction c pe of the activated complex potential energy diagrams time ms  $0\ 00\ 1\ 00\ 2\ 00\ 3\ 00\ 4\ 00\ 5\ 00\ 6\ 00\ 7\ 00\ 8\ 00\ 9\ 00\ 10\ 0$  potential energy kcal  $0\ 00\ 10\ 0$ 

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web use the potential energy diagram to answer the questions below 1 is the overall reaction as shown exothermic or endothermic 2 what is the activation energy for the forward reaction 3 what is the