Acids and Bases

Question 1:		Choose the correct answer for each of the following questions:					
1.1	The element present in all acids						
	A	oxygen	В	carbon	dioxide		
	C	nitrogen	D	hydre			
1.2	A sample of an acid in a beaker has a pH of 4. An alko. *n it. W. pH						
	the liquid in the beaker?						
	A	A decreases		stays the same			
	C	Increases	D	neutral			
1.3	Choose the incorrect combination:						
	A	HNH₄OH – Nitric acid.	В	H ₂ SO	sulphuric ac		
	C	HCl - Hydrochloric acid.	D		rbonic acid.		
1.4	The colour of blue litmus paper in an acid is						
	A	yellow	В	en			
	c	blue	r				
1.5	Which substance is always produced in eaction. ochloric and and lithium?						
	A	water		hydrogen ₈ .			
	C	oxygen gas			on dioxide gas		
1.6	Which	substance is always produ	in the re	en.	n hydrochloric acid and sodium		
	hydroxide?						
	A	water		hydrogen gas			
	C	oxygen gas		ecip	ecipitate		
1.7	Which formula repr						
	A	А КОН			KCI		
	C	CH ₃ OH		CH₃COOH			
1.8	Tell	od of adding an a a b	irop	by drop:			
		neutralisation	В	precipit	ation		
		eva-	D	decomp	position	[8]	
Quest.		ne genera reaction for an a			the following compounds:		
2.1	Base:						
2.2	Carbon	onate:					
2.3	Metal:			-@s	tudychamp.co).za	

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