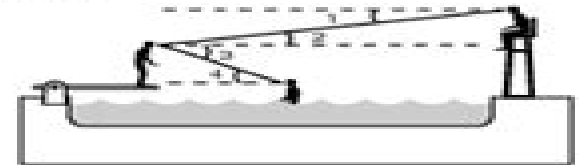


LESSON **8-4** **Practice A** **Angles of Elevation and Depression**

In Exercises 1 and 2, fill in the blanks to complete the definitions.

1. An angle of elevation is the angle formed by a _____ line and a line of sight to a point _____ the line.
2. An angle of _____ is the angle formed by a horizontal line and a line of sight to a point _____ the line.

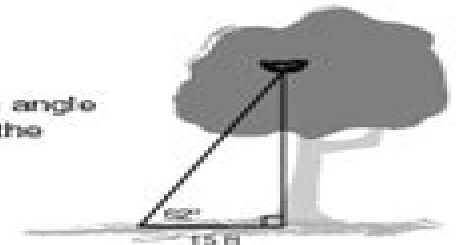
Ben is on the diving board at the neighborhood pool. Jenna is in the pool, and a lifeguard sits at her station on the opposite end of the pool. Classify each angle as an angle of elevation or an angle of depression.



3. $\angle 1$ _____
4. $\angle 2$ _____
5. $\angle 3$ _____
6. $\angle 4$ _____

Lisa sees a bird's nest high in a tree. She decides to use trigonometry to estimate how high the nest is.

7. Lisa walks 15 feet from the base of the tree. She measures an angle of elevation from the ground to the nest of 62° . Find how high the nest is above the ground, to the nearest foot.



8. Lisa spots the mother bird on a branch above the nest. She measures an angle of elevation to the bird of 67° . Find how high the mother bird is above the ground, to the nearest foot.

Zelda, a trapeze artist, stands on a 10-meter-high platform.

9. Zelda measures a 40° angle of depression to the base of the other platform. Find the distance between the bases of the platforms to the nearest tenth of a meter
10. Zelda's partner, Zev, is on the ground doing a safety check on the net. Zelda measures a 79° angle of depression to Zev. Find the distance to the nearest tenth of a meter from Zev to the base of Zelda's platform.



Angles Of Elevation And Depression Worksheet Lesson 8

4

Roman Wölfel



Angles Of Elevation And Depression Worksheet Lesson 8 4:

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web we interpret this as two molecules of hydrogen react with one molecule of oxygen to make two molecules of water the chemical equation is balanced as long as the coefficients are in the ratio 2 1 2 for instance this chemical equation is also balanced $100\text{H}_2 + 50\text{O}_2 \rightarrow 100\text{H}_2\text{O}$ $100\text{H}_2 + 50\text{O}_2 \rightarrow 100\text{H}_2\text{O}$

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web this experimental approach required the introduction of a new unit for amount of substances the mole which remains indispensable in modern chemical science the mole is an amount unit similar to familiar units like pair dozen gross etc it provides a specific measure of the number of atoms or molecules in a bulk sample of matter

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web the amount in moles of pb 3 o 4 to 3 significant figures iii the amount in moles of lead produced iv mass of lead produced to 3 significant figures b 35 61 g of tin reacts with exactly 42 60 g of chlorine cl 2 to form 78 21 g of tin iv chloride sncl 4 i calculate the number of moles of tin chlorine and tin chloride a r values

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