

# Calorimetry and Specific Heat Lab

## Part 1; Introduction

**Purpose-** The purpose of this lab is to investigate how a specific heat substance can be determined using a coffee cup calorimeter.

**Question-** Using a calorimeter, how can we figure out a specific heat?

**Hypothesis-** If the metal has a low heat, the metal can be a good material for cooking because it will not need as much heat to raise the temperature of the metal in order to cook food.

## Part 2; Materials

|                         |                                         |                          |                             |                  |
|-------------------------|-----------------------------------------|--------------------------|-----------------------------|------------------|
| Thermometer             | 250 mL beaker                           | 400 mL beaker            | Ring stand                  | Two buret clamps |
| Water                   | analytical balance                      | Test-tube holder         | Four test tubes             | Hot plate        |
| Lid for polystyrene cup | aluminum, copper, iron, and lead powder | 50 mL graduated cylinder | Two polystyrene coffee cups |                  |
|                         |                                         |                          |                             |                  |
|                         |                                         |                          |                             |                  |

**Step 1; Gather materials**

**Step 2; Put together the coffee cup calorimeter.**

- I Put one coffee cup in another and then put them in the 250ml beaker.
- I then Cut a hole in the center of the lid to hold our thermometer. Put the lid on the top cup.

**Step 3: Measure Masses of the Metals**

# Answer Sheet For Calorimetry Specific Heat

**L Towne**



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