

8. Buffer, passing through the, is shown within the Earth's upper. Buffer is composed and the composition of these all according to the.

12 samples of Buffer from a sampling were analysed to give the following composition of samples.

sample	composition (%)
1%	0.001
1%	0.01
1%	0.02

- (a) Define the term relative atomic mass.

The relative atomic mass of an element is the average mass of the atoms of the element, taking into account the relative abundance of the isotopes of the element.

- (b) Calculate the relative atomic mass of the sample of buffer.

Give your answer to three decimal places.

$$\left(\frac{1}{100} \times 1.001 \right) + \left(\frac{1}{100} \times 1.01 \right) + \left(\frac{1}{100} \times 1.02 \right)$$

1.011

answer = *1.011* (3) (3)

- (c) Some studies, on early fossil, indicate that organisms were made up of long particles which could be seen through microscopes, whereas some of the particles of subatomic particles in atoms and in ions.

Complete the table to show the number of subatomic particles in the ^{12}C atom and $^{12}\text{C}^{2+}$ ion.

	protons	neutrons	electrons
^{12}C	<i>6</i>	<i>6</i>	<i>6</i>
$^{12}\text{C}^{2+}$	<i>6</i>	<i>6</i>	<i>4</i>



Answer To F321 Chemistry May Series

Karin Nielsen-Saines



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