



Unit: Chemical Bonding Answer Key "Covalent Bonding" - WS #3

Directions: Please answer each question using your notes from today's class. ☺

1. What is the difference between a molecular formula, structural formula and an electron dot formula? Give an example of each.

A molecular formula gives the chemical make-up of one molecule. It does not show atom-to-atom connectivity. Example: C_3H_8

A structural formula uses lines instead of dots to show shared pairs of electrons.

Example: $H-H$

An electron dot formula uses dots instead of lines to show shared pairs of electrons.

Example: $H:H$

2. What is the difference between single, double, and triple covalent bonds? How many shared electrons are in each?

The difference between the types of bonds is: a single bond is two shared electrons, a double is four shared electrons, and a triple is six shared electrons.

3. Draw the electron dot and structural formulas for the following molecules.

Substance	Rough copy (work!)	Electron dot	Structural
a) Cl_2		<i>Note: The electron dot structures have dots in between each atom instead of lines.</i>	$:\ddot{Cl}-\ddot{Cl}:$
b) H_2S			$\begin{array}{c} :\ddot{S}: \\ \\ H \end{array} - H$
c) BrI			$:\ddot{I}-\ddot{Br}:$

Answers For Packet On Chemical Bonding

Gerhard Quinkert



Answers For Packet On Chemical Bonding:

Bonding, Structure and Solid-state Chemistry Marcus Frederick Charles Ladd, 2016 This book provides a study in Bonding Structure and Solid State Chemistry It is based on lecture courses given over several years but is not directed at any particular degree course Thus it will find a place in all years of first degree courses in both chemistry and those subjects for which chemistry forms a significant part It will also prepare readers for more intensive study in the title topics Pre knowledge is assumed in mathematics and physical sciences at about final year high school level Additional mathematical and other topics are presented where necessary as appendices so as not to disturb the flow of the main text The book is copiously illustrated including many stereoscopic diagrams with practical advice on correct viewing and colour illustrations A suite of computer programs some of which are interactive has been devised for the book and is available on line from the publisher's website global.oup.com/booksites/content/9780199670888 They are available for both 32 and 64 bit operating systems and are easily executed on a PC or laptop notes on their applications are provided Problems have been devised for each chapter and fully worked tutorial solutions are included After an introductory chapter the book presents a study based on the main interactive forces responsible for cohesion in the solid state of matter No classification is without some ambiguity but that chosen allows for a structured discussion over a wide range of compounds Each chapter includes worked examples on the study topics which together with the problems provided should ensure a thorough understanding of the textual material

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Music of Life Stanislav Tregub, 2020-08-08 Based on the Theory of Energy Harmony developed in the first two volumes of the series this book builds bridges between physical and life sciences It tackles fundamental questions that are considered the main riddles of biology How do the four elements of non living matter form the basic structures of living matter What is the physical mechanism that binds them into biochemical molecules The proposed model proceeds from the assumption that there are no special biophysical laws and the mechanism of energy interactions is universal for all types of matter The elements that form organic compounds have characteristics that allow them to create the Music of Life using this mechanism Starting from a detailed account of how notes of this music arise the book takes the reader step by step into the intricacies of the complex melodies

harmonies and rhythms of living matter By answering fundamental questions about non animate and animate Matter the author paves the way to answering fundamental questions about the Mind These questions are easy to formulate What is it What does it do Why does it do it How does it do it However they are not easily answerable from a physical perspective which is the only perspective science can take Many people argue that the mission is impossible and explain the Mind away by proclaiming that it is non physical and thus inaccessible for scientific study The author shows that the lack of explanation is only due to cognitive science remaining at the phenomenological level of study It only describes the observed physiological phenomena and their mental outcomes leaving an explanatory gap between them However as in physics causal explanation is the ultimate task in cognitive science We should uncover the mechanism that works in living systems and produces the mental phenomena we combine in the word Mind In this volume the author builds the foundation for the theory of the Mind developed in the following parts of the Symphony of Matter and Mind project

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Resources in Education ,1995

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