

Name: _____
Date: _____ Period: _____

Chemistry: *The Periodic Table*

Directions: Answer each of the following questions. You need not use complete sentences.

What is a period? How many are there in the periodic table?

A period is a row of elements on the periodic table, in which the elements are ordered one proton more each place from left to right. A period also often helps to represent an electron energy level and the order in which electrons are filled into their sublevels. There are 7 periods.

What is a group (also called a family)? How many are there in the periodic table?

Groups are the columns that are found top to bottom in the table. They show elements that have the same sublevel and number of electrons in their outer shell, valence electrons. Often elements in a group share any similar chemical properties because of this. There are 18 groups in the periodic table.

State the number of valence electrons in an atom of:

a. sulfur
6

b. calcium
2

c. chlorine
7

d. arsenic
5

Give the names and chemical symbols for the elements that correspond to these atomic numbers:

a. 10

Neon, Ne

b. 18

Argon, Ar

c. 36

Krypton, Kr

d. 90

Thorium, Th

List, by number, both the period and group of each of these elements.

	Symbol	Period	Group
a. beryllium	Be	2	2
b. iron	Fe	4	8
c. lead	Pb	6	14

Which of the following pairs of elements belong to the same period?

a. Na and Cl

b. Na and Li

c. Na and Cu

d. Na and Ne

Which of the following pairs of elements belong to the same group?

a. H and He

b. Li and Be

c. C and Pb

d. Ga and Ge

Would you expect strontium to be, chemically, more similar to calcium or rubidium and WHY?

Calcium, because they share a common amount of valence electrons. Valence electrons are the main contributor to chemical properties and how elements interact with other matter. Both strontium and calcium have two electrons on the the "s" sublevel, $5s^2$ and $4s^2$

What are the Group 1 elements called?

Answers Key To Chemistry Radioactivity Packet 15

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High Level Radioactive Waste Management ,1995 *Posture Hearings (NASA and FAA)* United States. Congress. House. Committee on Science and Technology,1980 **Radioactive Waste Management** ,1981 **Environmental Chemistry in the Lab** Ruth Ann Murphy,2022-08-31 Environmental Chemistry in the Lab presents a comprehensive approach to modern environmental chemistry laboratory instruction together with a complete experimental experience The laboratory experiments have an introduction for the students to read a pre lab for them to complete before coming to the lab a data sheet to complete during the lab and a post lab which would give them an opportunity to reinforce their understanding of the experiment completed Instructor resources include a list of all equipment and supplies needed for 24 students a lab preparation guide an answer key to all pre lab and post lab questions sample data for remote learners and a suggested rubric for grading the labs Additional features include Tested laboratory exercises with instructor resources for environmental science students Environmental calculations industrial regulation and environmental stewardship Classroom and remote exercises An excellent user friendly and thought provoking presentation which will appeal to students with little or no science background A qualitative approach to the chemistry behind many of our environmental issues today **Study Package for Indian Air Force Airmen Group Y (Non-Technical Trades) Exam with 3 Online Sets** Disha Experts,2019-08-12 Scientific and Technical Aerospace Reports ,1995 *Title List of Documents Made Publicly Available* ,1985 Study Package for Indian Air Force Airmen Group X & Y (Technical & Non Technical Trades) Exam with 3 Online Sets Disha Experts,2019-08-12 **Resources in Education** ,1997 **Budapest '92** ,1992 Key-words-in-context Title Index ,1962 **Nuclear Science Abstracts** ,1974 **Introduction to Chemical Exposure and Risk Assessment** W.Brock Neely,2020-04-28 Introduction to Chemical Exposure and Risk Assessment focuses on the principles involved in assessing the risks from chemical exposure These principles include the perception of risk an understanding of how numbers are handled and how chemicals affect health The book briefly describes the major sinks such as water and air where chemicals are introduced This is followed by a discussion on how concentrations are estimated and risk assessments are made A discussion of risk benefit analysis and a presentation of several case studies using the principles for assessing risks are also included Publications of the National Bureau of Standards, 1986 Catalog United States. National Bureau of Standards,1987 Publications United States. National Bureau of Standards,1987 **Publications of the National Institute of Standards and Technology ... Catalog** National Institute of Standards and Technology (U.S.),1987 **Chemistry, Energy and the Environment** César Sequeira,J B Moffat,2014-01-30 Focuses on key developments in the environmentally friendly production of energy and its conservation through an enhanced understanding of the chemical processes involved **Atomic Industry Reporter** ,1955 **Journal of Research of the National Bureau of Standards** United States. National Bureau of Standards,1975 Energy Research Abstracts ,1993 Semiannual with semiannual and

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In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Answers Key To Chemistry Radioactivity Packet 15**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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