Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

There are eight markers in a full set, but Flora and Frank each only have seven markers.
Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers?

Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?

Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the Covalent Bonds Gizmo, atoms form bonds in this way.

To begin, check that Fluorine is selected from the Select a substance menu. Click Play (**) to see the electrons orbiting the nucleus of each atom.



- The outermost electrons in each atom are called valence electrons. How many valence electrons does each fluorine atom have?
- Click Pause (III). Drag an electron from the left atom to the right atom. Click Play. What happens?
- Click Pause, drag an electron from the right atom to the left, and then click Play. What happens now?

Answers Ofexploration Guide Covalent Bonds

JR Anderson

Answers Of exploration Guide Covalent Bonds:

Unveiling the Energy of Verbal Artistry: An Mental Sojourn through Answers Ofexploration Guide Covalent Bonds

In a global inundated with monitors and the cacophony of quick interaction, the profound energy and mental resonance of verbal artistry often fade in to obscurity, eclipsed by the continuous assault of noise and distractions. However, nestled within the musical pages of **Answers Ofexploration Guide Covalent Bonds**, a captivating perform of literary beauty that pulses with fresh feelings, lies an unique journey waiting to be embarked upon. Penned with a virtuoso wordsmith, this enchanting opus books viewers on a mental odyssey, delicately revealing the latent potential and profound impact stuck within the delicate web of language. Within the heart-wrenching expanse of this evocative analysis, we can embark upon an introspective exploration of the book is central styles, dissect their charming publishing model, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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