

Buffers Keep the Balance

Data Table 1. Model Carbonate Blood Buffer

Table 1	Test Tube Number					
	1	2	3	4	5	6
Solution	Carbonic Acid (Reference)	Model Carbonate Blood Buffer	Water (Control)	Model Carbonate Blood Buffer	Water (Control)	Sodium Bicarbonate (Reference)
Bromothymol blue indicator color	yellow 6.0	greenish 7.2 blue	green 6.0	greenish 7.0 blue	green 6.0	blue
Initial pH value	6.0	7.2	6.0	7.0	6.0	8.0
Number of drops of HCl required to convert solutions 2 and 3 to yellow (acid reference color)	NA	16	5	NA	NA	NA
Number of drops of NaOH required to convert solutions 4 and 5 to blue (base reference color)	NA	NA	NA	2	2	NA

Data Table 2. Effect of HCl on Biological Phosphate Buffers

	Estimated pH (pH paper)	Universal Indicator Color			
		Initial	After 1 drop of HCl	After 5 drops of HCl	After 10 drops of HCl
Water (control)	6.0	6.0	6.0	NA	NA
Phosphate Buffer A	6.0	6.0	6.0	6.0	6.0
Phosphate Buffer B	6.4	6.4	6.4	6.0	6.0

Data Table 3. Effect of NaOH on Biological Phosphate Buffers

	Estimated pH (pH paper)	Universal Indicator Color			
		Initial	After 1 drop of NaOH	After 5 drops of NaOH	After 10 drops of HCl
Water (control)	6.0	6.0	7.0	NA	NA
Phosphate Buffer A	6.0	6.0	6.8	7.0	8.0
Phosphate Buffer B	6.4	6.4	7.0	7.0	7.0

Answers Buffers Keep The Balance Lab Answers

Mark Steyvers



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Biochemistry in the Lab Benjamin F. Lasseter, 2019-09-30 Most lab manuals assume a high level of knowledge among biochemistry students as well as a large amount of experience combining knowledge from separate scientific disciplines Biochemistry in the Lab A Manual for Undergraduates expects little more than basic chemistry It explains procedures clearly as well as giving a clear explanation of the theoretical reason for those steps Key Features Presents a comprehensive approach to modern biochemistry laboratory teaching together with a complete experimental experience Includes chemical biology as its foundation teaching readers experimental methods specific to the field Provides instructor experiments that are easy to prepare and execute at comparatively low cost Supersedes existing older texts with information that is adjusted to modern experimental biochemistry Is written by an expert in the field This textbook presents a foundational approach to modern biochemistry laboratory teaching together with a complete experimental experience from protein purification and characterization to advanced analytical techniques It has modules to help instructors present the techniques used in a time critical manner as well as several modules to study protein chemistry including gel techniques enzymology crystal growth unfolding studies and fluorescence It proceeds from the simplest and most important techniques to the most difficult and specialized ones It offers instructors experiments that are easy to prepare and execute at comparatively low cost

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Genetic Engineering News ,2006 **The Commercial Development of New Products** Michael G. Duerr,1986 **Freshwater and Marine Aquarium** ,1996 *Life* ,1953 *Graphics Interface 2008* Lyn Bartram,Chris Shaw,2008-12-03 This volume with contributions by top graphics researchers and professionals is devoted to computer graphics interactive systems and human computer interaction The paper contributions are from talks given at the annual Graphics Interface conference now in its 34th year Graphics Interface occupies a unique niche among conferences in that it seeks to both combine and bridge research topics in and across these areas Beginning in 1969 as the Canadian Man Computer Communications Seminar CMCCS it is the oldest regularly scheduled computer graphics and human computer interaction conference This year Graphics Interface was held May 28 30 2008 in Windsor Ontario Graphics Interface GI 2008

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