Balancing Chemical Equations Worksheet

1.
$$2 H_2 + 0_2 \rightarrow 2 H_2O$$

2.
$$N_2 + 3 H_2 \rightarrow 2 NH_3$$

4.
$$2 N_2 + 0_2 \rightarrow 2 N_20$$

5.
$$2 \text{ HgO} \rightarrow 2 \text{ Hg} + 0$$

6.
$$6 CO_2 + 6 H_2O \rightarrow C_6H_{12}O_6 + 6 O_2$$

9.
$$\underline{2}$$
 Na + $\underline{2}$ H₂O \Rightarrow $\underline{2}$ NaOH + $\underline{}$ H₂

10.
$$_{-2}$$
 $H_3PO_4 \rightarrow _{---}$ $H_4P_2O_7 + _{---}$ H_2O

12. ____
$$CO_2$$
 + ___ Z $NH_3 \rightarrow ____ $OC(NH_2)_2$ + ____ $H_2O$$

13.
$$4 \text{ Si}_2\text{H}_3 + 7 \text{ O}_2 \rightarrow 8 \text{ SiO}_2 + 6 \text{ H}_2\text{O}_3$$

14. _2_Al(OH)₃ + _3_H₂SO₄
$$\Rightarrow$$
 ___Al₂(SO₄)₃ + _6_H₂O

16. ____ Fe₂(SO₄)₃ + ___6_ KOH
$$\rightarrow$$
 ___3_ K₂SO₄ + __2_ Fe(OH)₃

17.
$$2 C_7H_6O_2 + 15 O_2 \rightarrow 14 CO_2 + 5 H_2O$$

19. _____ FeS₂ + _____ O₂
$$\rightarrow$$
 _____ Fe₂O₃ + _____ SO₂

21. ____Fe₂O₃ + __3_H₂
$$\rightarrow$$
 __2_Fe + __3_H₂O

22. ____Na₂CO₃ + ___ HCl
$$\rightarrow$$
 __ NaCl + ____H₂O + ____CO₂

24. ____
$$C_7H_{16} + _//_ O_2 \rightarrow _//_ CO_2 + _//__ H_{2}O$$

25.
$$P_4 + 5 O_2 \rightarrow 7 P_2O_5$$

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