

Name: Dominick A. Grede

Date: _____

Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

- There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers? Share

- Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?

Share 2

Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the *Covalent Bonds* Gizmo, atoms form bonds in this way.

To begin, check that **Fluorine** is selected from the **Select a substance** menu. Click **Play** () to see the electrons orbiting the nucleus of each atom.



- The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have? 7

- Click **Pause** (). Drag an electron from the left atom to the right atom. Click **Play**.

What happens? it orbits both atoms

- Click **Pause**, drag an electron from the right atom to the left, and then click **Play**.

What happens now? it orbits both atoms

Answer Key To Student Exploration Covalent Bonds

EW Minium



Answer Key To Student Exploration Covalent Bonds:

Exploration of the Structure of Atom Chandan Sengupta, First Publication April 2019 Revised Publication October 2022
Second Revised Edition July 2023 Third Revised Edition December 2024 Published by Chandan Senguta Printed by IECIT
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Even More Brain-powered Science Thomas O'Brien, 2011 The third of Thomas O'Brien's books designed for 5th-12 grade science teachers Even More Brain Powered Science uses questions and inquiry oriented discrepant events or experiments or demonstrations in which the outcomes are not what students expect or to dispute misconceptions and challenge students to think about discuss and examine the real outcomes of the experiments O'Brien has developed interactive activities many of which use inexpensive materials to engage the natural curiosity of both teachers and students and create new levels of scientific understanding

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