

Types of Chemical Reaction Worksheet

Balance the reactions **1 to 6** and indicate which type of chemical reaction (synthesis, decomposition, single-displacement, double-displacement or combustion) is being represented:

1. $2 \text{NaBr} + \text{Ca}(\text{OH})_2 \rightarrow \text{CaBr}_2 + 2 \text{NaOH}$ Reaction Type : single displacement
2. $2 \text{NH}_3 + \text{H}_2\text{SO}_4 \rightarrow 3 (\text{NH}_4)_2\text{SO}_4$ Reaction Type : synthesis
3. $4 \text{C}_3\text{H}_8\text{O} + 27 \text{O}_2 \rightarrow 20 \text{CO}_2 + 18 \text{H}_2\text{O}$ Reaction Type : combustion
4. $3 \text{Pb} + 2 \text{H}_3\text{PO}_4 \rightarrow 3 \text{H}_2 + \text{Pb}_3(\text{PO}_4)_2$ Reaction Type : single displacement
5. $\text{Li}_3\text{N} + 3 \text{NH}_4\text{NO}_3 \rightarrow 3 \text{LiNO}_3 + (\text{NH}_4)_3\text{N}$ Reaction Type : double displacement
6. $\text{HBr} + 3 \text{Al}(\text{OH})_3 \rightarrow \text{H}_2\text{O} + 3 \text{AlBr}_3$ Reaction Type : double displacement

Indicate which type of chemical reaction (synthesis, decomposition, single-displacement, double-displacement or combustion) is being represented in 7 to 20.

7. $\text{Na}_3\text{PO}_4 + 3 \text{KOH} \rightarrow 3 \text{NaOH} + \text{K}_3\text{PO}_4$ Reaction Type **double displacement**
8. $\text{MgCl}_2 + \text{Li}_2\text{CO}_3 \rightarrow \text{MgCO}_3 + 2 \text{LiCl}$ Reaction Type **double displacement**
9. $\text{C}_8\text{H}_{12} + 9 \text{O}_2 \rightarrow 6 \text{CO}_2 + 6 \text{H}_2\text{O}$ Reaction Type **combustion**
10. $\text{Pb} + \text{FeSO}_4 \rightarrow \text{PbSO}_4 + \text{Fe}$ Reaction Type **single displacement**
11. $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$ Reaction Type **decomposition**
12. $\text{P}_4 + 3 \text{O}_2 \rightarrow 2 \text{P}_2\text{O}_3$ Reaction Type **synthesis**
13. $2 \text{RbNO}_3 + \text{BeF}_2 \rightarrow \text{Be}(\text{NO}_3)_2 + 2 \text{RbF}$ Reaction Type **double displacement**
14. $2 \text{AgNO}_3 + \text{Cu} \rightarrow \text{Cu}(\text{NO}_3)_2 + 2 \text{Ag}$ Reaction Type **single displacement**
15. $\text{C}_3\text{H}_8\text{O} + 4 \text{O}_2 \rightarrow 3 \text{CO}_2 + 3 \text{H}_2\text{O}$ Reaction Type **combustion**
16. $2 \text{C}_2\text{H}_5 + \text{Fe} \rightarrow \text{Fe}(\text{C}_2\text{H}_5)_2$ Reaction Type **synthesis**
17. $\text{SeCl}_6 + \text{O}_2 \rightarrow \text{SeO}_2 + 3 \text{Cl}_2$ Reaction Type **single displacement**
18. $2 \text{MgI}_2 + \text{Mn}(\text{SO}_3)_2 \rightarrow 2 \text{MgSO}_3 + \text{MnI}_4$ Reaction Type **double displacement**
19. $\text{O}_3 \rightarrow \text{O} + \text{O}_2$ Reaction Type **decomposition**
20. $2 \text{NO}_2 \rightarrow 2 \text{O}_2 + \text{N}_2$ Reaction Type **decomposition**

Answers For Chemical Reactions

V. K. Sally



Answers For Chemical Reactions:

Chemical Reactions, Materials and Particles Gill Murphy, 2006-03 Specials are a teaching resource designed for KS3 students whose literacy skills are considerably lower than their age. These books have an older format to counteract the simple text and cover various topics. They include activities, visuals, and assessment sheets as well as teacher pages.

Regents Exams and Answers: Chemistry--Physical Setting Revised Edition Barron's Educational Series, Albert Tarendash, 2021-01-05 Barron's Regents Exams and Answers: Chemistry provides essential practice for students taking the Chemistry Regents, including actual recently administered exams and thorough answer explanations for all questions. This book features eight actual administered Regents Chemistry exams so students can get familiar with the test. Thorough explanations for all answers, self-analysis charts to help identify strengths and weaknesses, test-taking techniques and strategies, a detailed outline of all major topics tested on this exam, a glossary of important terms to know for test day.

Examination Questions and Answers in Basic Anatomy and Physiology Martin Caon, 2020-08-03 This third edition provides 2900 multiple choice questions on human anatomy and physiology and some biophysical science, separated into 20 chapters and 68 categories. In addition, there are 64 essay topics. The answer to each question is accompanied by an explanation. Each chapter has an introduction to set the scene for the questions to come. However, not all possible information is provided within these introductions, so an Anatomy and Physiology textbook is an indispensable aid to understanding the answers. The textbook offers a more holistic approach to the subjects of anatomy and physiology by also including biomechanics, biophysics, and biochemistry. The questions have been used in end-of-semester examinations for undergraduate anatomy and physiology courses and, as such, reflect the focus of these particular courses and are pitched at this level to challenge students that are beginning their training in anatomy and physiology. The question and answer combinations are intended for use by teachers to select questions for their next examinations and by students when studying for an upcoming test. Students enrolled in the courses for which these questions were written include nursing, midwifery, paramedic, physiotherapy, occupational therapy, nutrition, and dietetics, health sciences, exercise science, and students taking an anatomy and physiology course as an elective.

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notes are tailored to the latest syllabus of CBSE Class VII Science exams making them a valuable resource for exam preparation

A New Approach to ICSE Chemistry for Class IX (A.Y. 2023-24) Onward V. K. Sally, 2023-05-20

The book A New Approach to ICSE Chemistry for Class IX has been written in accordance with the latest syllabus prescribed by the Council for Indian School Certificate Examination New Delhi. The book is divided into five distinct parts:

1. The first part constitutes chapters on The Languages of chemistry, Chemical Changes and Reactions, and Water, to give an idea to the students about the diversity of matter.
2. The second part constitutes chapters on Atomic Structure and Chemical Bonding and The Periodic Table. This part will help the students to understand unity in diversity of matter. It further explains the rules required for reaction between various forms of matter.
3. The third part constitutes chapter on Study of the First Element Hydrogen and Study of Gas Laws. This will familiarise the students with element hydrogen and their chemical properties. It further explains the behaviour of gases under changes of temperature and pressure and also provides explanation in terms of molecular motion.
4. The fourth part deals with Atmospheric Pollution.
5. The fifth part deals with Practical Chemistry.

The salient features of the book:

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- All definitions and other important points are highlighted.
- Diagrams are made simple and more informative.
- In the chapter on the language of chemistry, a new method has been introduced for balancing chemical equations. This method is based on atomic numbers, which is a fundamental property of the elements.
- A chapter on Practical Chemistry is added to help students for laboratory work.
- Questions have been added chapter wise under the heading Question Bank for the benefit of students.
- The solution can be accessed through QR Code given at the end of each chapter.
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