

Post-Lab Questions

1. From the experiment, which reactant is the limiting reactant? Which one is the reactant in excess? How do you determine which one is the limiting reactant?
2. If you weigh the watch glass then write your name on it with the wax pencil, how will that affect your percent yield?
3. What factors could cause the percent yield of a product to be less than 100%?
4. What factors could cause the percent yield of a product to be more than 100%?
5. Balance the following reaction: $\text{C}_3\text{H}_8 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
 - a. 4.30 g of C_3H_8 reacts with 4.30 g of O_2 . Which one is the limiting reagent, and which one is in excess? Explain your reasoning.
 - b. Based on the limiting reagent, what are the theoretical yields of CO_2 and H_2O ?
 - c. You perform the experiment and produce 2.67 g of CO_2 and 0.79 g of H_2O . What are your percent yields for each product?

Answers Key To Chemistry 1211 Lab Manual

Sebastian Brünink



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In a world inundated with displays and the cacophony of instantaneous conversation, the profound power and mental resonance of verbal artistry usually disappear into obscurity, eclipsed by the regular barrage of noise and distractions. Yet, nestled within the musical pages of **Answers Key To Chemistry 1211 Lab Manual**, a charming function of literary elegance that impulses with fresh thoughts, lies an wonderful trip waiting to be embarked upon. Composed by way of a virtuoso wordsmith, this mesmerizing opus books visitors on a psychological odyssey, softly exposing the latent possible and profound impact stuck within the complicated web of language. Within the heart-wrenching expanse of this evocative analysis, we can embark upon an introspective exploration of the book is key styles, dissect their captivating publishing model, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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