

FACTORY WIRING CODES

NOTE: • All wire colors are viewed as looking at the front of the plug with the wires exiting the rear.
• Use these wiring codes as a guide. Your vehicle might differ.

BLACK 16-SPKER INPUT *

- | | |
|---------|---------------------------|
| 1 = L2+ | 8 = R1+ |
| 2 = L2- | 9 = R1- |
| 3 = R2+ | 10 = R2- (Pass Interlock) |
| 4 = R2- | 11 = R3+ |
| 5 = L1+ | 12 = L3+ |
| 6 = L1- | 13 = L3- |
| 7 = R1+ | 14 = Battery +10V |

*Tapped to stock



1996-97 VEHICLES

- | |
|------------------|
| 1 = Ground |
| 2 = Ground |
| 3 = Battery +10V |
| 4 = Ignition |
| 5 = Ignition |
| 6 = Ground |
| 7 = Ignition |
| 8 = Battery +10V |

GRAY PLUG 8-PIN



BLACK 16-PIN

- | | |
|-----------------------------|------------------------|
| 1 = Left Indicator (L) +10V | 9 = Ignition |
| 2 = Ignition | 10 = Memory/Default on |
| 3 = Red (Lamp) | 11 = Ignition |
| 4 = Red (Lamp) | 12 = Ignition |
| 5 = Red (Lamp) | 13 = Red (Lamp) |
| 6 = Red (Lamp) | 14 = Ignition |
| 7 = Red (Lamp) | 15 = Red (Lamp) |
| 8 = Red (Lamp) | 16 = Red (Lamp) |

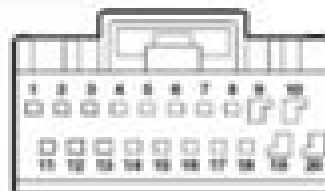
Red (Lamp) = Substrate
Not for aftermarket use!



1998-UP VEHICLES

POWER AND SPEAKER

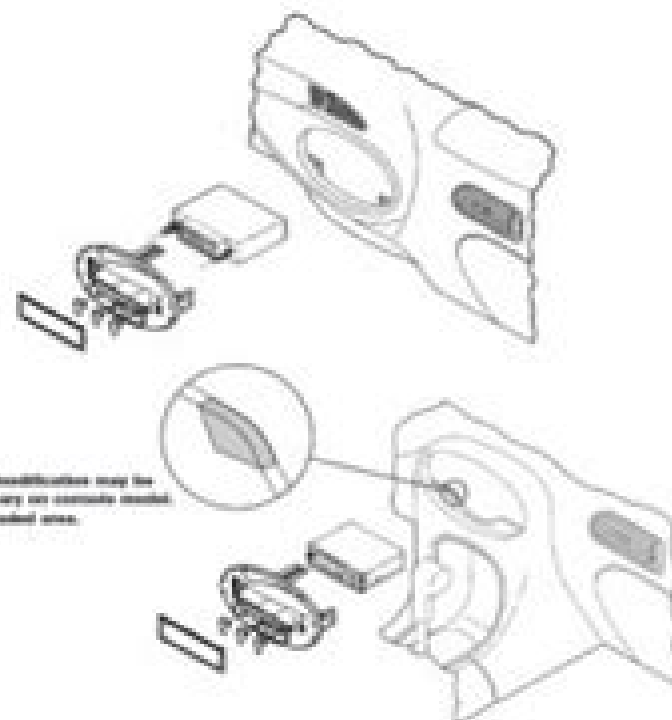
- | |
|--|
| 1 = Black/Red = Red (Lamp) |
| 2 = Blue/White = Red (Lamp) |
| 3 = Black = Red (Lamp) |
| 4 = Green/White = Indicator or Indicator |
| 5 = Black/Red = +10V Accessory |
| 6 = Ignition |
| 7 = Light Green/Black = Red (Lamp) |
| 8 = Ignition |
| 9 = Light Blue/Black = Red (Lamp) |
| 10 = Black (Lamp) (Lamp) = +10V (Chassis Ground) |
| 11 = Brown/Red (Lamp) = Red (Lamp) |
| 12 = Ignition = Red (Lamp) |
| 13 = Ignition |
| 14 = Red (Lamp) = Indicator Switch (Red) |
| 15 = Ignition |
| 16 = Red/White = Red (Lamp) |
| 17 = Red/White = Red (Lamp) |
| 18 = Red/White = Red (Lamp) |
| 19 = Light Blue/Black = Red (Lamp) |
| 20 = Black/Red (Lamp) = Red (Lamp) |
| 21 = White = +10V Constant |



SCOSCHE

FD134030

IN-DASH INSTALLATION KIT FOR 1996-99 FORD TAURUS 1996-99 MERCURY SABLE WITH ROTARY CLIMATE CONTROLS



Minor modification may be necessary on console model.
See shaded area.

LIABILITY DISCLAIMER

This instruction booklet is based on carefully researched data and research of automobile dash assemblies, wire harnesses and information pertaining to installation of this kit (FD134030) on 1996-99 Ford Taurus and Mercury Sable Vehicles. Scosche Industries, Inc. can not be held responsible for discrepancies/inconsistencies that may occur due to the automobile during the installation of components while using this booklet.

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2008 Ford Fusion Radio Wiring Diagram

E.I. Electrical Press

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the 'E.I. Electrical Press' text.

2008 Ford Fusion Radio Wiring Diagram:

Adopting the Song of Expression: An Mental Symphony within **2008 Ford Fusion Radio Wiring Diagram**

In some sort of used by screens and the ceaseless chatter of quick connection, the melodic splendor and mental symphony produced by the prepared term usually disappear into the backdrop, eclipsed by the relentless sound and disruptions that permeate our lives. However, situated within the pages of **2008 Ford Fusion Radio Wiring Diagram** a charming fictional treasure brimming with natural feelings, lies an immersive symphony waiting to be embraced. Constructed by an outstanding composer of language, this charming masterpiece conducts readers on a mental journey, well unraveling the hidden melodies and profound influence resonating within each cautiously crafted phrase. Within the depths with this emotional analysis, we shall discover the book is main harmonies, analyze their enthralling writing design, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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web these ions are called spectator ions because they do not participate in the actual reaction canceling the spectator ions gives the net ionic equation which shows only those species that participate in the chemical reaction $2\text{Ag}^+ \text{aq}$ $\text{Cr}_2\text{O}_7^{2-} \text{aq}$ $\rightarrow$ $2\text{Cr}^{3+} \text{aq}$ $7\text{H}_2\text{O}$ l $2\text{H}^+ \text{aq}$

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web jan 29 2020 the ionic form of this reaction is $2\text{Na}^+\text{aq} + 2\text{Cl}^-\text{aq} + \text{Cu}^{2+}\text{aq} + \text{SO}_4^{2-}\text{aq} \rightarrow 2\text{Na}^+\text{aq} + \text{SO}_4^{2-}\text{aq} + \text{CuCl}_2\text{s}$ the sodium ions and sulfate ion are the spectator ions in this reaction they appear unchanged in both the product and reactant side of the equation

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web mar 11 2014 solution ionic equation $\text{H}^+\text{aq} + \text{Cl}^-\text{aq} + \text{Na}^+\text{aq} + \text{HCO}_3^-\text{aq} \rightarrow \text{Na}^+\text{aq} + \text{Cl}^-\text{aq} + \text{H}_2\text{O}\text{l} + \text{CO}_2\text{g}$ the spectator ions are Cl^- and Na^+
 net ionic equation $\text{H}^+\text{aq} + \text{HCO}_3^-\text{aq} \rightarrow \text{H}_2\text{O}\text{l} + \text{CO}_2\text{g}$
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web the six Na^+aq ions and the six Cl^-aq ions that appear on both sides of the equation are spectator ions that can be canceled to give the net ionic equation $3\text{Ba}^{2+}\text{aq} + 2\text{PO}_4^{3-}\text{aq} \rightarrow \text{Ba}_3(\text{PO}_4)_2\text{s}$

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web video answer you know spectator ions are the ions present on the both sides of the equation indicate cancels out in the total ionic equations so you know in a case chemical reactions are described with chemical equations which symbolically present the starting materials end products

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water $\text{H}^+\text{aq} + \text{OH}^-\text{aq} \rightarrow \text{H}_2\text{O}\text{l}$ the spectator ions in KOHaq and HNO_3aq are K^+aq and NO_3^-aq how to identify and list the

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anions an ion with a net charge of zero is electrically neutral

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web aug 8 2022 a spectator ion is an ion that does not take part in the chemical reaction and is found in solution both before and after the reaction the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction

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