

## Chapter 1 - Chemical Foundations

1. Consider the element indium, atomic number 49, atomic mass 114.8 g. The nucleus of an atom of indium-112 contains:

- a. 49 protons, 63 neutrons, 49 electrons.
- b. 49 protons, 49 neutrons.
- c. 49 protons, 49 alpha particles.
- d. 49 protons, 63 neutrons.
- e. 49 protons, 112 neutrons.

ANSWER:

d

POINTS:

1

2. A hypothetical element consists of two isotopes of masses 78.95 amu and 80.95 amu with abundances of 32.9% and 67.1%, respectively. What is the average atomic mass of this element?

- a. 79.95 amu
- b. 78.95 amu
- c. 79.6 amu
- d. 80.3 amu
- e. 80.95 amu

ANSWER:

d

POINTS:

1

3. Naturally occurring element X exists in three isotopic forms: X-28 (27.977 amu, 92.23% abundance), X-29 (28.976 amu, 4.67% abundance), and X-30 (29.974 amu, 3.10% abundance). Calculate the atomic weight of X.

- a. 28.09 amu
- b. 48.63 amu
- c. 27.16 amu
- d. 28.97 amu
- e. 86.93 amu

ANSWER:

a

POINTS:

1

4. The atomic mass of rhenium is 186.2. Given that 37.1% of natural rhenium is rhenium-185, what is the other stable isotope?

- a.  $^{183}_{75}\text{Re}$
- b.  $^{187}_{75}\text{Re}$
- c.  $^{186}_{75}\text{Re}$
- d.  $^{181}_{75}\text{Re}$
- e.  $^{190}_{75}\text{Re}$

ANSWER:

b

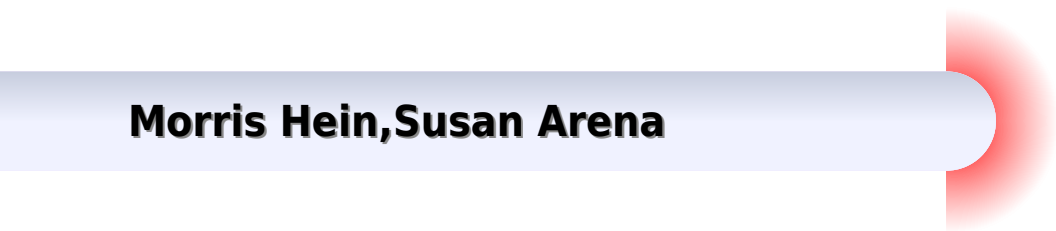
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5. Naturally occurring element X exists in three isotopic forms: X-28 (27.977 amu, 92.23% abundance), X-29

# Answers For Chapter One Chemical Foundations Exercises

**Morris Hein, Susan Arena**



**Answers For Chapter One Chemical Foundations Exercises:**

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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Answers For Chapter One Chemical Foundations Exercises**," an enthralling opus penned by a highly acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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