

Net Ionic Equations Worksheet

Write balanced chemical equations for dissolving the following compounds in water:

1. Solid sodium chloride



2. Solid barium nitrate



3. Solid magnesium hydroxide



4. Gaseous hydrochloric acid



5. Solid calcium phosphate



6. Solid iron (III) sulfide



7. Gaseous carbon dioxide (*Hint: Carbon dioxide will form carbonic acid first then decomposes into ions*)



Write balanced chemical equations and then net ionic equations for the following reactions:

8. Aqueous sodium sulfate and aqueous barium chloride



9. Aqueous lithium bromide and aqueous silver nitrate



10. Aqueous lead II acetate and aqueous sodium carbonate



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Rahmawati, Peter Charles Taylor, 2019-06-07 This conference proceedings focuses on enabling science and mathematics practitioners and citizens to respond to the pressing challenges of global competitiveness and sustainable development by transforming research and teaching of science and mathematics The proceedings consist of 82 papers presented at the Science and Mathematics International Conference SMIC 2018 organised by the Faculty of Mathematics and Natural Sciences Universitas Negeri Jakarta Indonesia The proceedings are organised in four parts Science Science Education Mathematics and Mathematics Education The papers contribute to our understanding of important contemporary issues in science especially nanotechnology materials and environmental science science education in particular environmental sustainability STEM and STEAM education 21st century skills technology education and green chemistry and mathematics and its application in statistics computer science and mathematics education **Net Ionic Equations** Lifeliqe, 2019 This lesson plan covers the various ways to depict chemical equations using net ionic equations **Salt Effects on rates of ionic reactions in solutions** Yael Weissmann, 1967 **Ionic Reactions and Equilibria** , 1967

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