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Disseminating Behavioral Research Annette K. Griffith, Tyler C. Ré, 2023-12-09 Dissemination is a key component of the research process While several fields have developed resources dedicated to training and supporting their scientists and practitioners as they are encouraged to disseminate within their fields and to the larger public audience there has been a lack of formal guidance for dissemination within the behavioral sciences Disseminating Behavioral Science aims to fill that gap providing guidance across modalities for topics ranging from the peer review process to conference presentation to nontraditional avenues for dissemination The contents of this edited text divided across six sections serves as a roadmap for students junior researchers and senior scholars The first section includes types of academic scholarship types of dissemination and strategies to ensure ethical dissemination The second reviews traditional publication preparation including tips for the writing process and key components to include in each section of a scientific manuscript Section Three explores publication within a traditional peer reviewed journal Section Four outlines additional strategies to get research publicly recognized through conference presentations social media and popular media sources and white and grey literature Sections Five and Six offer a consolidated glossary of all key terms in the text and combined reference list

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physical to calculate the percent composition we need to know the masses of c h and o in a known mass of $C_9H_8O_4$ it is
convenient to consider 1 mol of $C_9H_8O_4$ and use its physical science if8767 answer key heat

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compound one must know the molar mass of the elements and the compound in order to get percent composition for instance
the percent composition of oxygen in water is 89

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the chemical formula of calcium hydroxide is given as Ca(OH)_2 based on the molecular formula of calcium hydroxide
the formula mass can be calculated by getting the sum of the molar mass of each of the constituent

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16 07 g zn 20 00 g 100 80 35 zn o 3 93 g o 20 00 g 100 19 65