

“All My Sons” – Notes on Act 3

1. How does Miller use Larry’s letter to add dramatic tension to the final act of the play? Consider its use on pages 75, 78 and 79.
 - Ann is using the letter as a last resort to prove that Larry is dead. It answers an important question that has remained unanswered throughout the play, revealing that Larry is dead, and that this is a direct result of Joe’s war profiteering.
 - The fact that Miller introduces the letter but doesn’t reveal its contents to the audience for several pages builds up dramatic tension, as it increases our desires to find out about its contents. This tension is heightened by the reaction of various characters:
 - Kate reads it: “[A long, low moan comes from mother’s throat]”
 - “My God, my God”
 - Ann tries to hand the letter to Chris to read, but Kate tries to stop her, terrified of him finding out what is in it.
 - Kate then pleads with Chris not to read it to Joe “[Pleading with her whole soul] Don’t tell him.”
2. Consider Keller’s speech on page 79 in relation to the play as a whole.
 - a) Why has Miller given the play the title “All My Sons”?
 - It focuses on this quotation from Keller’s final speech in the play where he faces up to the responsibility he has for society as a whole – it is not enough for him to look after the interests of his own sons (Larry and Chris), he should also protect the interests of society as a whole.
 - b) Why does Miller include the play’s title in Keller’s speech on this page?
 - This shows that Keller has learned the moral lesson that Miller is trying to draw our attention to in the play – that we all have to take responsibility for our wider society.
3. In what ways have the following characters changed from the way they were represented at the start of the play?

You may find it helpful to refer to the start of the play in your answer.

- a) Chris
 - At the start of the play Chris idolised Joe “Isn’t he a great guy”, but at the end of the play he forces his dad to hand himself over to the police for his involvement in war profiteering, an act that leads Joe to kill himself. “Now if I look at him, all I’m able to do is cry”.
 - Chris has now faced up to Joe’s guilt in shipping the machine heads after years of fooling himself into thinking he was innocent “I suspected my father and I did nothing about it” p76
- b) Kate (“Mother”)
 - Kate has been forced to accept Larry’s death, right up until Ann hands her Larry’s suicide letter. “How did he die? You’re lying to me. If you know, how did he die?” Even late in act three, Kate still continues to insist that Larry is alive, demanding that Ann prove he is dead when she suggests it.
- c) Joe (“Keller”)
 - Keller has been forced to accept that his war profiteering has directly led to Larry’s death, as seen in his “all my sons speech”.

All My Sons Act 3 Answers

Albert Gibson, Robert McLean



All My Sons Act 3 Answers:

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web the percent yield is calculated as follows $\text{percent yield} = \frac{\text{actual yield}}{\text{theoretical yield}} \times 100$ percent yield theoretical yield actual yield 100 based on this definition we would expect a percent yield

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ch₃oh can be produced by the following reaction $2\text{H}_2 + \text{CO} \rightarrow \text{CH}_3\text{OH}$ calculate the theoretical yield of ch₃oh if 68.5 g of co is reacted with 8.6 g of h₂

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web limiting reactant and percent yield practice name 1 consider the following reaction $\text{NH}_4\text{NO}_3 + \text{Na}_3\text{PO}_4 \rightarrow \text{NH}_4\text{PO}_4 + \text{NaNO}_3$ which reactant is limiting assuming we started with 30.0 grams of ammonium nitrate and 50.0 grams of sodium phosphate what is the

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web limiting reagents and percentage yield worksheet 1 consider the reaction $\text{I}_2\text{O}_5 + 5\text{CO} \rightarrow 5\text{CO}_2 + \text{I}_2$ a 80.0 grams of iodine v oxide i₂o₅ reacts with 28.0 grams of carbon monoxide co determine the mass of iodine i₂ which could be produced 80 g i₂o₅ 1 mol i₂o₅ 1 mol i₂ xs 1 333.8 g i₂o₅ 1 mol i₂o₅ 28 g co 1 mol co

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web jun 2 2020 the percent yield of a reaction is the ratio of the actual yield to the theoretical yield multiplied by 100 to give a percentage text percent yield text actual yield g over text theoretical yield g times 100 label 3 7 3