

SOLUBILITY CURVES

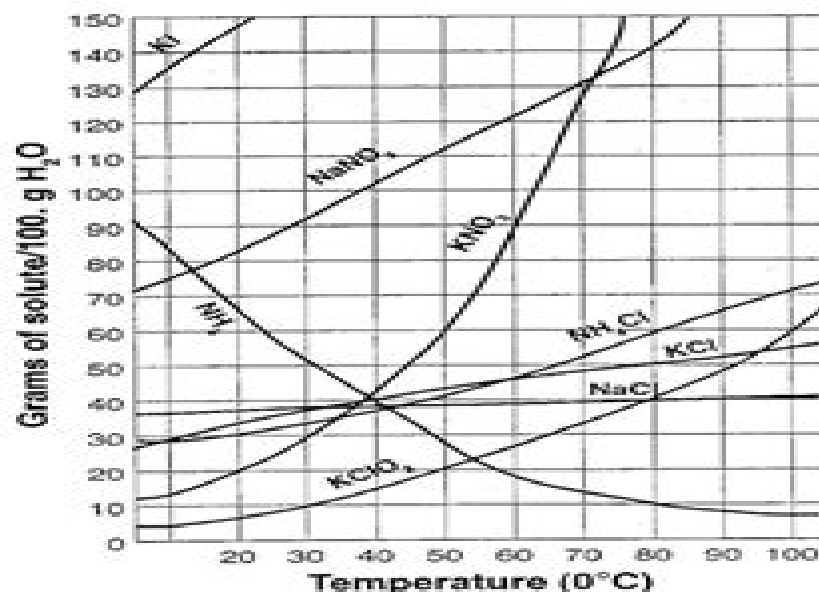
Name _____

Answer the following questions based on the solubility curve below.

1. Which salt is least soluble in water at 20° C? _____
2. How many grams of potassium chloride can be dissolved in 200 g of water at 80° C?

3. At 40° C, how much potassium nitrate can be dissolved in 300 g of water? _____
4. Which salt shows the least change in solubility from 0° – 100° C?

5. At 30° C, 90 g of sodium nitrate is dissolved in 100 g of water. Is this solution saturated, unsaturated or supersaturated?



6. A saturated solution of potassium chlorate is formed from one hundred grams of water. If the saturated solution is cooled from 80° C to 50° C, how many grams of precipitate are formed? _____
7. What compound shows a decrease in solubility from 0° to 100° C? _____
8. Which salt is most soluble at 10° C? _____
9. Which salt is least soluble at 50° C? _____
10. Which salt is least soluble at 90° C? _____

Answer Key To Solubility Curve 1

Cambridge International Examinations



Answer Key To Solubility Curve 1:

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liquid solutions vapors or even in solid state Notwithstanding its inherently high complexity the crystallization process is part of our everyday lives from ice making in our homes to the most state of the art chemical and electronic industry In this book our purpose was to present new insights to the reader as well as crucial and very useful information for researchers working in this field while simultaneously creating a comprehensive text about crystallization processes which may serve as a starting point for people with different backgrounds *The Pearson Guide to Objective Chemistry for the AIEEE* Singhal

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