

PHYSICAL: Kinetic Molecular Theory

Learning Objectives

- At standard atmospheric conditions, all gases have approximately the same kinetic energy, but are made of different molecules and are made up of different numbers of particles. (The particles of the gases are different sizes.)
- At standard atmospheric conditions, all the particles of the same gas have the same kinetic energy.
- At standard atmospheric conditions, the average speed of the particles of a gas is inversely proportional to the square root of its molar mass.

Model 1: Representation of Gases in Different Phases

The following figure illustrates the representation of particles in three phases: solid, liquid, and gas.



Figure 1: Representation of particles in different phases of matter

Model 1 represents the particles in different phases of matter. The particles in the solid and liquid phases are closely packed, while the particles in the gaseous phase are widely spaced.

Key Questions

1. Which one of the three representations of particles in different phases of matter is most similar to the particles in the solid phase? Which one is most similar to the particles in the liquid phase? Which one is most similar to the particles in the gaseous phase? Justify your answers.

Table 1: Representation of particles in different phases of matter

	Solid	Liquid	Gaseous
Which one of the three representations of particles in different phases of matter is most similar to the particles in the solid phase? Justify your answer.	The particles in the solid phase are closely packed and arranged in a regular pattern. The particles in the liquid phase are closely packed but disordered. The particles in the gaseous phase are widely spaced and disordered.	The particles in the liquid phase are closely packed but disordered. The particles in the solid phase are closely packed and arranged in a regular pattern. The particles in the gaseous phase are widely spaced and disordered.	The particles in the gaseous phase are widely spaced and disordered. The particles in the solid phase are closely packed and arranged in a regular pattern. The particles in the liquid phase are closely packed but disordered.
Which one of the three representations of particles in different phases of matter is most similar to the particles in the liquid phase? Justify your answer.	The particles in the solid phase are closely packed and arranged in a regular pattern. The particles in the liquid phase are closely packed but disordered. The particles in the gaseous phase are widely spaced and disordered.	The particles in the liquid phase are closely packed but disordered. The particles in the solid phase are closely packed and arranged in a regular pattern. The particles in the gaseous phase are widely spaced and disordered.	The particles in the gaseous phase are widely spaced and disordered. The particles in the solid phase are closely packed and arranged in a regular pattern. The particles in the liquid phase are closely packed but disordered.
Which one of the three representations of particles in different phases of matter is most similar to the particles in the gaseous phase? Justify your answer.	The particles in the solid phase are closely packed and arranged in a regular pattern. The particles in the liquid phase are closely packed but disordered. The particles in the gaseous phase are widely spaced and disordered.	The particles in the liquid phase are closely packed but disordered. The particles in the solid phase are closely packed and arranged in a regular pattern. The particles in the gaseous phase are widely spaced and disordered.	The particles in the gaseous phase are widely spaced and disordered. The particles in the solid phase are closely packed and arranged in a regular pattern. The particles in the liquid phase are closely packed but disordered.

Answers Kinetic Molecular Theory Pogil

Stephen G. Brush



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