

Chemistry Unit 7 Review

701 – Inquiry Lab

- Define...
 - Direct relationship: Two variables increase/decrease together
 - Indirect relationship: One variable increases, the other decreases
- Identify whether the listed variables have a direct or indirect relationship to each other.
 - Moles and volume: Direct
 - Pressure and volume: Indirect
 - Temperature and pressure: Direct
 - Moles and pressure: Direct
- Using the Ideal Gas Law equation, list 3 ways you could increase the volume of a balloon.
Increase the moles, increase the temperature, decrease the pressure (more so, the pressure drops as a result of greater volume)
- Using the Ideal Gas Law equation, list 3 ways you could decrease the pressure within a syringe.
Decrease the moles, decrease the temperature, increase the volume
- If the number of moles in a container is doubled, by how much should the pressure increase (assuming everything else is held constant)?
The pressure should also double

702 – Gas Laws Equation

- What equation should you use when you are given just one set of variables?
 $PV = nRT$
- You have a 4.5 L canister of O_2 that you store in a room that is $20^\circ C$. The pressure of this tank is 13000 kPa. How many moles of O_2 are in this canister?
24 moles

Answer Key Chemistry U7 Ws4

B Lingard



Answer Key Chemistry U7 Ws4:

The book delves into Answer Key Chemistry U7 Ws4. Answer Key Chemistry U7 Ws4 is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Answer Key Chemistry U7 Ws4, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Answer Key Chemistry U7 Ws4
- Chapter 2: Essential Elements of Answer Key Chemistry U7 Ws4
- Chapter 3: Answer Key Chemistry U7 Ws4 in Everyday Life
- Chapter 4: Answer Key Chemistry U7 Ws4 in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Answer Key Chemistry U7 Ws4. The first chapter will explore what Answer Key Chemistry U7 Ws4 is, why Answer Key Chemistry U7 Ws4 is vital, and how to effectively learn about Answer Key Chemistry U7 Ws4.
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5. In chapter 4, this book will scrutinize the relevance of Answer Key Chemistry U7 Ws4 in specific contexts. The fourth chapter will explore how Answer Key Chemistry U7 Ws4 is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Answer Key Chemistry U7 Ws4. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Answer Key Chemistry U7 Ws4.

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