#### Ionic and Covalent Naming Practice Answers

For each of the following questions, determine whether the compound is ionic or covalent and name it appropriately.

1)	Na <sub>2</sub> CO <sub>3</sub>	sodium carbonate	
2)	P4O10	tetraphosphorus decaoxide	
3)	N <sub>2</sub> O <sub>5</sub>	dinitrogen pentoxide	
4)	FeSO <sub>4</sub>	iron (II) sulfate	
5)	SiO <sub>2</sub>	silicon dioxide	
6)	FeCl <sub>3</sub>	iron (III) chloride	
7)	CoBr <sub>2</sub>	cobalt (II) bromide	
8)	B <sub>2</sub> H <sub>4</sub>	diboron tetrahydride	
9)	co	carbon monoxide	
10)	PCI <sub>3</sub>	phosphorus trichloride	

For each of the following questions, determine whether the compound is ionic or covalent and write the appropriate formula for it.

11)	dinitrogen trioxide	N2O3
12)	nitrogen dioxide	NO <sub>2</sub>
13)	sulfur pentoxide	SO <sub>s</sub>
14)	lithium acetate	LiC <sub>2</sub> H <sub>3</sub> O <sub>2</sub>
15)	phosphorus trifluoride	PF <sub>3</sub>
16)	vanadium (V) oxide	V <sub>2</sub> O <sub>5</sub>
17)	aluminum hydroxide	AI(OH)3
18)	zinc sulfide	ZnS
19)	silicon tetrafluoride	SiF4
20)	silver phosphate	Ag <sub>3</sub> PO <sub>4</sub>

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**Holt Rinehart & Winston** 

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