

Why?

When working in chemistry, it is often convenient to write a chemical in symbols. For example we might write down the substance table salt as NaCl . In talking about chemistry however, it is a bit tacky to say "en-ay see-dell" when we want to refer to a substance. Also, in formal writing we should use the name of the compound rather than its symbols. Therefore we need to learn how to say the proper names of ionic substances.

1	H ⁺									
2	Li ⁺	Be ²⁺						N ³⁻	O ²⁻	F ⁻
3	Na ⁺	Mg ²⁺	Transition elements				Al ³⁺	P ³⁻	S ²⁻	Cl ⁻
4	K ⁺	Ca ²⁺	Fe ²⁺ Fe ³⁺	Ni ²⁺ Ni ³⁺	Cu ⁺ Cu ²⁺	Zn ²⁺				Br ⁻
5	Rb ⁺	Sr ²⁺			Ag ²⁺			Sn ²⁺ Sn ⁴⁺		I ⁻
6		Ba ²⁺				Hg ₂ ²⁺ Hg ²⁺		Pb ²⁺ Pb ⁴⁺		
	Cations						Anions			

4. 1. Based on the information in Model 1:
- Identify three elements that form only one cation.
 $\text{Li, Na, K, Rb, Cs, Mg, Ca, Sr, Ba, Ag, Zn, Al}$ (H according to this sheet, but technically not)
 - Identify three elements that form only one anion.
 $\text{N, P, O, S, F, Cl, Br, I}$
 - Identify three elements that form more than one cation.
 $\text{Fe, Ni, Cu, Hg, Sn, Pb}$
 - In what region of the periodic table are these "multiple ion" elements usually located?
Center, transition elements (metals)
1. 2. Consider the ions of potassium (K) and sulfur (S). Write chemical formulas for all possible ionic compounds involving these ions, using the simplest ratio(s) of potassium (K) and sulfur (S). Keep in mind that the sum of the charges in an ionic compound must equal zero.
 K_2S
2. 3. Consider the ions of iron (Fe) and sulfur (S). Write chemical formulas for all possible ionic compounds involving these ions, using the simplest ratio(s) of iron (Fe) and sulfur (S). Keep in mind that the sum of the charges in an ionic compound must equal zero.



Answer Key For Making Models Of Compounds

Max Rechtman



Answer Key For Making Models Of Compounds:

TEXT BOOK OF COMPUTER AIDED DRUG DESIGN Valapa Anusha, Lalbihari Barik, Prashant Gupta, Dr Pichika Mallikarjuna Rao, Mak Kit-Kay, 2025-05-30 The Text Book of Computer Aided Drug Design is a comprehensive guide covering modern techniques used in computational drug discovery It begins with an introduction to Computer Aided Drug Design CADD highlighting its history fundamental principles and wide ranging applications The book then delves into Quantitative Structure Activity Relationships QSAR explaining basics the evolution of QSAR methodologies and the importance of physicochemical parameters like electronic lipophilicity and steric effects Both experimental and theoretical approaches for parameter determination are detailed Further it elaborates on Hansch and Free Wilson analysis deriving 2D QSAR equations and advanced 3D QSAR approaches along with contour map interpretation A dedicated section discusses the crucial role of molecular modeling and quantum mechanics in drug design It contrasts global minimum energy conformations with bioactive conformations and thoroughly explains rigid flexible and extra precision molecular docking techniques The text also explores enzyme targets such as DHFR HMG CoA reductase HIV protease and cholinesterases emphasizing the design of inhibitors Another highlight is the prediction of ADMET properties essential for successful drug candidates De novo drug design is explored with focus on receptor enzyme interactions cavity predictions and fragment based approaches Techniques like homology modeling and generation of 3D protein structures are covered to support structure based drug design The final chapters are dedicated to pharmacophore mapping and virtual screening methods Readers learn about pharmacophore identification conformational search techniques in silico drug design strategies and both similarity based and structure based virtual screening approaches Rich in theory and practical approaches this book serves as an essential resource for pharmacy medicinal chemistry and computational biology students It bridges fundamental concepts with advanced drug discovery techniques It is ideal for both beginners seeking a strong foundation and researchers aiming for advanced applications Comprehensive examples models and updated techniques make it highly relevant to current pharmaceutical research and industry needs *Scientifica Essentials*, 2005 Bring your science lessons to life with Scientifica Providing just the right proportion of reading versus doing these engaging resources are differentiated to support and challenge pupils of varying abilities

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Atoms, Molecules & Elements: What Are Compounds? Gr. 5-8

George Graybill,2015-10-01 This is the chapter slice What Are Compounds from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way

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