

Activity A:**Ions**Get the Gizmo ready:

- Click **Reset**.
- Check that sodium and chlorine are still selected.



Introduction: Some of the particles that make up atoms have an electrical charge. Electrons are negatively charged, while protons are positively charged. Particles with opposite charges (+ and –) attract, while particles with the same charge (+ and + or – and –) repel.

Question: What happens when atoms gain or lose electrons?

1. **Count:** Electrons move around the nucleus of atoms in specific shells, shown by the rings around the atoms in the Gizmo. The first ring holds two electrons, and the second holds eight. (If you like, you can hide the inner electrons by selecting **Hide inner electrons**.)

- A. Observe the sodium and chlorine atoms. How many electrons are there total in each atom?

Sodium: 11

Chlorine: 17

- B. Each atom is neutrally charged, which means that each atom has the same number of protons and electrons. Based on this, how many protons are in each atom?

Sodium: 11

Chlorine: 17

2. **Transfer:** Drag electrons from one atom to another until you have a charged sodium cation. The cation is shown in the Gizmo. How many electrons does the cation have now?

Sodium: 10

Chlorine: 17

3. **Calculate:** Each electron has a charge of -1 , and each proton has a charge of $+1$. The net charge is the charge of all electrons, subtracting the number of electrons from the number of protons. What is the net charge of the sodium cation?

- A. What is the charge of each electron? Sodium: -1 Chlorine: -1

Now use these charges to check. Does the charged atom's net charge equal 0?

- B. Is each ion neutral? Explain.

Yes, the net charge is zero because the number of electrons and protons is equal.

Now click on the plus sign on the right side of the Gizmo to check.

4. **Electron sharing:** What happens if electrons between the two atoms in the Gizmo are shared?

Nothing happens because the net charge is still zero.

Answer Key Of Gizmo Additive Colours

Patrick Vollmar



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Table of Contents Answer Key Of Gizmo Additive Colours

1. Understanding the eBook Answer Key Of Gizmo Additive Colours
 - The Rise of Digital Reading Answer Key Of Gizmo Additive Colours
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key Of Gizmo Additive Colours
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Key Of Gizmo Additive Colours
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key Of Gizmo Additive Colours
 - Personalized Recommendations
 - Answer Key Of Gizmo Additive Colours User Reviews and Ratings
 - Answer Key Of Gizmo Additive Colours and Bestseller Lists

5. Accessing Answer Key Of Gizmo Additive Colours Free and Paid eBooks
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 - Answer Key Of Gizmo Additive Colours eBook Subscription Services
 - Answer Key Of Gizmo Additive Colours Budget-Friendly Options
6. Navigating Answer Key Of Gizmo Additive Colours eBook Formats
 - ePub, PDF, MOBI, and More
 - Answer Key Of Gizmo Additive Colours Compatibility with Devices
 - Answer Key Of Gizmo Additive Colours Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer Key Of Gizmo Additive Colours
 - Highlighting and Note-Taking Answer Key Of Gizmo Additive Colours
 - Interactive Elements Answer Key Of Gizmo Additive Colours
8. Staying Engaged with Answer Key Of Gizmo Additive Colours
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer Key Of Gizmo Additive Colours
9. Balancing eBooks and Physical Books Answer Key Of Gizmo Additive Colours
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Key Of Gizmo Additive Colours
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answer Key Of Gizmo Additive Colours
 - Setting Reading Goals Answer Key Of Gizmo Additive Colours
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answer Key Of Gizmo Additive Colours
 - Fact-Checking eBook Content of Answer Key Of Gizmo Additive Colours
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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