Name:

mashupmath

Balancing Chemical Equations

Directions: Balance each of the following chemical equations.

- ___ H₂PO₄ + ___ KOH → ___ K₂PO₄ + ___ H₂O 13
- __ K + __ B₂O₃ → __ K₂O + __ B 20
- ___ HCl + ___ NaOH → ___ NaCl + ___ H₂O 3)
- ___ Na + ___ NaNO₃ -> ___ Na₂O + ___ N₂ 40
- $_C + _S_0 \rightarrow _CS_2$ 50
- ___ Na + ___ O₂ -> ___ Na₂O 6)
- $N_2 + O_2 \rightarrow N_2O_5$ 7)
- $_{\text{max}}$ H₃PO₄ + $_{\text{max}}$ Mg(OH)₂ \rightarrow $_{\text{max}}$ Mg₃(PO₄)₂ + $_{\text{max}}$ H₂O 8)
- ___ NaOH + ___ H₂CO₃ → ___ Na₂CO₃ + ___ H₂O 9)
- ___ KOH + ___ HBr → ___ KBr + ___ H₂O 10)
- __ Na + __ O₂ → __ Na₂O 11)
- ___ Al (OH)₃ + ___ H₂CO₃ → ___ Al₂(CO₃)₃ + ___ H₂O 12)
- $Al + S_n \rightarrow Al_2S_1$ 13)
- ___ Cs + ___ N₂ -> ___ Cs₂N 14)
- __ Mg + __ Cl₂ → __ MgCl₂ 15)
- 16) ___ Rb + ___ RbNO₁ → ___ Rb₂O + ___ N₂
- 17) __C₆H₆ + __O₂ → __CO₂ + __H₂O
- 18) ___ N₂ + ___ H₂ → ___ NH₃

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John A. White, Kellie S.
Grasman, Kenneth E. Case, Kim LaScola
Needy, David B. Pratt

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