# Identifying the 5 Types of Chemical Reactions

Identify each of the reactions below as one of the 5 types of Reactions (Rxn):

- Synthesis Reaction
- Decomposition Reaction
- Single Replacement Reaction
- Double Replacement Reaction
- Combustion Reaction

# Reactions (identify pattern below)

Name of Reaction Type

- 1.  $2 \underbrace{Na}_{B} + \underbrace{Cl_{2}}_{A} \Rightarrow 2 \underbrace{NaCl}_{A}_{B}$  (A & B combine) Ex: synthesis
- Pb + FeSO<sub>4</sub> → PbSO<sub>4</sub> + Fe
- P<sub>4</sub> + 3 O<sub>2</sub> → 2 P<sub>2</sub>O<sub>3</sub>
- 2 NO₂ → 2 O₂ + N₂
- Na<sub>3</sub>PO<sub>4</sub> + 3 KOH → 3 NaOH + K<sub>3</sub>PO<sub>4</sub>
- C<sub>3</sub>H<sub>6</sub>O + 4 O<sub>2</sub> → 3 CO<sub>2</sub> + 3 H<sub>2</sub>O
- MgCl<sub>2</sub> + Li<sub>2</sub>CO<sub>3</sub> → MgCO<sub>3</sub> + 2 LiCl
- C<sub>6</sub>H<sub>12</sub> + 9 O<sub>2</sub> → 6 CO<sub>2</sub> + 6 H<sub>2</sub>O
- CaCO<sub>3</sub> → CaO + CO<sub>2</sub>
- 2 AgNO<sub>3</sub> + Cu → Cu(NO<sub>3</sub>)<sub>2</sub> + 2 Ag
- 2H₂ + O₂ → 2H₂O

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#### **Table of Contents Answer Classifying Chemical Reaction**

- 1. Understanding the eBook Answer Classifying Chemical Reaction
  - The Rise of Digital Reading Answer Classifying Chemical Reaction
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Answer Classifying Chemical Reaction
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Answer Classifying Chemical Reaction
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Answer Classifying Chemical Reaction
  - Personalized Recommendations
  - Answer Classifying Chemical Reaction User Reviews and Ratings
  - Answer Classifying Chemical Reaction and Bestseller Lists
- 5. Accessing Answer Classifying Chemical Reaction Free and Paid eBooks

- Answer Classifying Chemical Reaction Public Domain eBooks
- Answer Classifying Chemical Reaction eBook Subscription Services
- Answer Classifying Chemical Reaction Budget-Friendly Options
- 6. Navigating Answer Classifying Chemical Reaction eBook Formats
  - o ePub, PDF, MOBI, and More
  - Answer Classifying Chemical Reaction Compatibility with Devices
  - Answer Classifying Chemical Reaction Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Answer Classifying Chemical Reaction
  - Highlighting and Note-Taking Answer Classifying Chemical Reaction
  - Interactive Elements Answer Classifying Chemical Reaction
- 8. Staying Engaged with Answer Classifying Chemical Reaction
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Answer Classifying Chemical Reaction
- 9. Balancing eBooks and Physical Books Answer Classifying Chemical Reaction
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Answer Classifying Chemical Reaction
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Answer Classifying Chemical Reaction
  - Setting Reading Goals Answer Classifying Chemical Reaction
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answer Classifying Chemical Reaction
  - Fact-Checking eBook Content of Answer Classifying Chemical Reaction
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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