

Chemistry Worksheets Class 11 on Chapter 4 Chemical Bonding and Molecular Structure - Set 3

Q-1: Which of the following possesses an expanded octet?

- a) IF_7
- b) SF_6
- c) BrF_5
- d) All of the above

Q-2: Strongly electronegative element B contrasts with strongly electropositive element A. Both are divalent. The resulting compound would be

- a) $\text{A} \rightarrow \text{B}$
- b) $\text{A}^{2+}\text{B}^{2-}$
- c) $\text{A}^{2-}\text{B}^{2+}$
- d) $\text{A}-\text{B}$

Q-3: Which of the following compounds shows the highest lattice energy?

- a) CsCl
- b) KCl
- c) RbCl
- d) NaCl

Q-4: Which one of the following molecules is formed by p-p overlapping?

- a) I_2
- b) HBr
- c) H_2S
- d) PH_3

Q-5: Which of the following H-bonds has the greatest impact on a molecule's physical characteristics?

- a) Intramolecular H-bond
- b) Intermolecular H-bond
- c) Both of the above
- d) All H-bonds

Q-6: State the crucial conditions that must be met for a molecule to undergo hybridisation.

Q-7: Describe why CH_4 has a tetrahedral geometry rather than a square planar geometry with a carbon atom in the centre and four H atoms at each corner.

Answer Key For Holt Science Chemical Bonding

Xiang Xie



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Unveiling the Magic of Words: A Review of "**Answer Key For Holt Science Chemical Bonding**"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Answer Key For Holt Science Chemical Bonding**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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