

Chapter 9 Review

- 1) What does stoichiometry mean? "measuring elements"
- 2) Before performing a stoichiometry problem, you must first do what with the chemical equation? **balance it**
- 3) The coefficients in a chemical equation represent what? **moles**
- 4) In a chemical equation, what relationships are shown? **molar ratios**
- 5) Solve the following mole-mole conversions using this equation:



- a) If 3.5 moles of sodium bicarbonate are decomposed, how many moles of carbon dioxide are produced?

$$3.5 \text{ mol NaHCO}_3 \times \frac{1 \text{ mol CO}_2}{2 \text{ mol NaHCO}_3} = 1.8 \text{ mol CO}_2$$

- b) If 1.29 moles of water are produced by the decomposition of sodium bicarbonate, how many moles of sodium carbonate are produced?

$$1.29 \text{ mol H}_2\text{O} \times \frac{1 \text{ mol Na}_2\text{CO}_3}{1 \text{ mol H}_2\text{O}} = 1.29 \text{ mol Na}_2\text{CO}_3$$

- c) How many moles of sodium bicarbonate are required to produce 19.5 moles of carbon dioxide?

$$19.5 \text{ mol CO}_2 \times \frac{2 \text{ mol NaHCO}_3}{1 \text{ mol CO}_2} = 39.0 \text{ mol NaHCO}_3$$

- 6) Use the equation in # 5 to solve the following mass-mass conversions.

- a) If given 10.5 grams of sodium bicarbonate, how many grams of sodium carbonate can be produced?

$$10.5 \text{ g NaHCO}_3 \times \frac{1 \text{ mol NaHCO}_3}{84.01 \text{ g NaHCO}_3} \times \frac{1 \text{ mol Na}_2\text{CO}_3}{2 \text{ mol NaHCO}_3} \times \frac{105.99 \text{ g Na}_2\text{CO}_3}{1 \text{ mol Na}_2\text{CO}_3} = 6.62 \text{ g Na}_2\text{CO}_3$$

- b) How many grams of sodium bicarbonate would be required to produce 9.0 grams of water?

$$9.0 \text{ g H}_2\text{O} \times \frac{1 \text{ mol H}_2\text{O}}{18.02 \text{ g H}_2\text{O}} \times \frac{2 \text{ mol NaHCO}_3}{1 \text{ mol H}_2\text{O}} \times \frac{84.01 \text{ g NaHCO}_3}{1 \text{ mol NaHCO}_3} = 84 \text{ g NaHCO}_3$$

- 7) What conversion factor (mole ratio) would you use to solve for moles of sulfur formed from sulfur dioxide in the reaction



Answer Key To Chemistry B Moles Packet

Christian G. Meyer



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web a baseball game lasts for nine innings while softball lasts for seven innings there are three outs per half inning teams rotate batting and fielding each half inning tied games may be left as ties or be played until a winner is found known as extra innings

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web comparison of baseball and softball a comparison of the differences between baseball and softball can be made as softball is directly descended from baseball an observer of one game would find the other very similar but there are several important rule differences fastpitch softball is more popular in competitive leagues

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web rules of softball each team consists of 9 players and teams can be of mixed gender a game lasts for 7 innings and is split into two sections the top and bottom of the innings each team bats once in each innings before the sides switch the fielding team has a pitcher catcher a player on first base second base third base three deep

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web oct 13 2023 the fundamentals of softball are the same as those of baseball batting and fielding strategy are similar but softball is played on a much smaller area and a game is only seven innings long the regulation playing field for softball

includes a diamond shaped area with 60 foot 18 3 metre baselines

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web 5 min by olympics com baseball softball getty images baseball softball is one of the five additional sports proposed by the la28 organising committee for inclusion at the olympic games los angeles 2028 pending approval from the international olympic committee ioc at their session in mumbai october 2023

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web softball pitchers must use an underhand motion to pitch throwing the ball from under their arms baseball pitchers may use any method though overhand throws are most commonly used number of innings per game softball games are made up of seven innings while baseballers must play nine per game