

Radioactivity and Half-Lives Review Worksheet

- What is a half-life?
The **time** it takes for half of the material you started with to decay.
- If we start with 400 atoms of a radioactive substance, how many would remain after one half-life? **200** After 2 half-lives? **100** After 3 half-lives? **50** After 4 half-lives? **25**
- If we start with 16 grams of a substance, how much will remain after 3 half-lives?
 $16 \rightarrow 8 \rightarrow 4 \rightarrow 2$ **2 grams** after 3 half-lives.
- A sample of carbon-14 has been decaying for 22,920 years and is now 35.0 grams. What was the size of the original sample? (The half-life of carbon-14 is 5,730 years)
 $22920/5730 = 4$ half lives *You will need to work backwards*
 $35 \rightarrow 70 \rightarrow 140 \rightarrow 280 \rightarrow$ **560. grams**
- How old is a skeleton sample if the current amount of carbon in the bones is 3.125%? (Assume you started with 100%)
 $100\% \rightarrow 50 \rightarrow 25 \rightarrow 12.5 \rightarrow 6.25 \rightarrow 3.125$ 5 half lives $\times 5,730 =$ **28,650 years**
- If you started with 120 grams of a radioactive substance, and now you have 15 grams, how many half-lives have passed?
 $120 \rightarrow 60 \rightarrow 30 \rightarrow 15$ **3 half-lives**
- The half-life of isotope X is 2.0 years. How many years would it take for a 4.0 mg sample of isotope X to decay and have only 0.50 mg of it remain?
 $4.0 \rightarrow 2.0 \rightarrow 1.0 \rightarrow 0.50$ 3 half lives $\times 2$ years = **6.0 years**
- The half-life of Po-218 is three minutes. How much of a 200 atom sample will remain after 15 minutes?
 $15 \text{ minutes}/3 = 5$ half lives $200 \rightarrow 100 \rightarrow 50 \rightarrow 25 \rightarrow 12.5 \rightarrow$ **6.25 atoms**
- The half-life of Au-198 is 2.69 days. How long does it take a 180g sample to decay to 1/8 its original mass?
 $1 \rightarrow \frac{1}{2} \rightarrow \frac{1}{4} \rightarrow \frac{1}{8}$ 3 half lives $\times 2.69$ days = **8.07 days**
- What is the half-life of a radioactive sample if 100.0 grams of it decays to 12.5 grams in 24.3 hours?
 $100 \rightarrow 50 \rightarrow 25 \rightarrow 12.5$ $24.3/3$ half lives = **8.10 hours**
- How long will it take a sample of ^{131}I to decay to 12.5% assuming its half-life is 8.07 days?

Answer Key To Half Lives

G Thomas



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