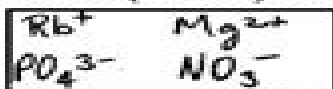


19. An aqueous solution of rubidium phosphate (Rb_3PO_4) is mixed with an aqueous solution of magnesium nitrate and a precipitate forms.

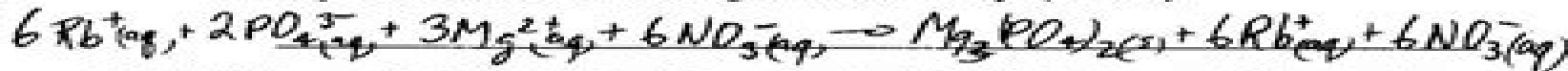
a) Draw an (ion-box) with the four ions involved with this reaction. (1 mark)



b) Write a *balanced formula equation* for this reaction. Include all subscripts and coefficients. (2 marks)



c) Write a balanced, *complete (total) ionic equation* for this reaction. Include all subscripts, coefficients and ion charges where necessary. (2 marks)



d) Write a *net ionic equation* for this reaction. Include all subscripts, coefficients and ion charges where necessary. (2 marks)



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stains bullet matching and drug lab evidence reveal about whodunit These mysteries are at the core of this teacher resource book which meets the unique needs of high school chemistry classes in a highly memorable way The book makes forensic evidence the foundation of a series of eight hands on week long labs As you weave the labs throughout the year and students solve the case the narrative provides vivid lessons in why chemistry concepts are relevant and how they connect All chapters include case information specific to each performance assessment and highlight the related national standards and chemistry content Chapters provide Teacher guides to help you set up Student performance assessments A suspect file to introduce the characters and new information about their relationships to the case Samples of student work that has been previously assessed and that serves as an answer key for you Grading rubrics Using Forensics in Chemistry as your guide you will gain the confidence to use inquiry based strategies and performance based assessments with a complex chemistry curriculum Your students may gain an interest in chemistry that rivals their fascination with Bones and CSI [Resources in Education](#),1998

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