

- 4 a Calcium ions are 2+ and chloride ions are 1-, so the formula is CaCl₂.
 - b Silver ions are 1+ and nitrate ions are 1-, so the formula is AgNO₃.
 - c Copper(s) is Cu²⁺ and phosphate is PO₄³⁻. For a neutral compound you need three copper ions and two phosphate ions, making charges of +6 and -6. Therefore, the formula is Qu₂(PO₄)₃₋.
 - d Aluminium is 3+ and oxide is 2-, so aluminium oxide is Al₂O₃.
 - There are two ways of working out formulae of ionic compounds. One way is to make sure that the total positive charge equals the total negative charge (see part c). The other way, if the charges are not the same, is to swap them round. Copper is 2+ and phosphate 3-, so there are three copper ions and two phosphate ions in the formula.
- 5 a As there is one manganese ion to each carbonate ion, the charges have the same numerical value. Carbonate ions are 2-, so the manganese ions in this compound must be 2+.
 - b The charge on two vanadium ions must equal the charge on three sulfate ions. Each sulfate is -2, so three sulfate ions have a total charge of -6. This means that two vanadium ions have a total charge of +6, so each has a charge of +3.
- 6 a molar mass of Ca(OH)₂ = 40.1 + 2 × (16.0 + 1.0) = 74.1 g mol⁻¹
 - b molar mass of $Al_2(SO_4)_1 = (2 \times 27.0) + 3 \times [32.1 + (4 \times 16.0)] = 342.3 \text{ g mol}^{-2}$
 - c molar mass of FeSO₄.7H₂O = $55.8 + 32.1 + (4 \times 16.0) + 7 \times (2.0 + 16.0) = 277.9 \text{ gmol}^{-1}$
 - ② In calculating molar mass, make sure that you use the relative atomic masses (the larger numbers), not the atomic numbers, from the periodic table. Molar mass has the units of g mol⁻¹. Relative molecular mass has no units. The numbers are the same.
- 7 The balanced equations are:
 - a $P_4 + 5O_2 \rightarrow P_4O_{23}$ or $2P_2O_5$
 - Phosphorus(v) oxide exists as the dimer P₂O₁₀, but 2P₂O₃ on the right of the equation would be acceptable.
 - b 4AI + 3O₂ → 2AI₂O₂
 - c $2Mg(NO_3)_2 \rightarrow 2MgO + 4NO_2 + O_3$
 - In part c, balance the oxygen atoms last. Remember that the number of oxygen atoms on the right-hand side of the equation must be an even number, because Mg(NO₂)₂ contains an even number of oxygen atoms.
 - d 2LiOH + H₂SO₄ → Li₂SO₄ + 2H₂O
 - $e^{-}2Fe^{3+} + 2f^{-} \rightarrow 2Fe^{2+} + I_{-}$
 - Make sure that this equation balances for charge. Having one Fe³⁺ ion on the left-hand side and one Fe³⁺ on the right-hand side balances the number of Felions, but the charge on the left would not equal the charge on the right.
 - $f = 2C_0H_{18} + 25O_2 \rightarrow 16CO_2 + 18H_2O$ or $C_0H_{18} + 12\frac{1}{2}O_2 \rightarrow 8CO_2 + 9H_2O$
 - The only source of oxygen on the left-hand side of the equation is O₂, so the number of oxygen atoms on the right-hand side must be an even number. Therefore, there must be an even number of H₂O molecules on the right. This requires two C₂H₁₂ molecules on the left.
- 8 For equations a-c, the simplest way is to realise that the component ions of the compound combine to give the precipitate. All other ions are spectators.

Answer Key For Teaching Transparency

Terry D. Warfield, Jerry J. Weygandt, Donald E. Kieso

Answer Key For Teaching Transparency:

Resources in Education,2000 Transparent Design in Higher Education Teaching and Leadership Mary-Ann Winkelmes, Allison Boye, Suzanne Tapp, 2023-07-03 This book offers a comprehensive guide to the Transparency in Learning and Teaching TILT framework that has convincingly demonstrated that implementation increases retention and improved outcomes for all students Its premise is simple to make learning processes explicit and equitably accessible for all students Transparent instruction involves faculty student discussion about several important aspects of academic work before students undertake that work making explicit the purpose of the work the knowledge that will be gained and its utility in students lives beyond college explaining the tasks involved the expected criteria and providing multiple examples of real world work applications of the specific academic discipline The simple change of making objective and methods explicit that faculty recognize as consistent with their teaching goals creates substantial benefits for students and demonstrably increases such predictors of college students success as academic confidence sense of belonging in college self awareness of skill development and persistence This guide presents a brief history of TILT summarizes both past and current research on its impact on learning and describes the three part Transparency Framework of purposes tasks and criteria The three sections of the book in turn demonstrate why and how transparent instruction works suggesting strategies for instructors who wish to adopt it describing how educational developers and teaching centers have adopted the Framework and concluding with examples of how several institutions have used the Framework to connect the daily work of faculty with the learning goals The Transparent Teacher Trent Kaufman, Emily that departments programs and institutions aim to demonstrate Grimm, 2013-02-15 A new paradigm for teachers to open their classrooms and cultivate deep professional learning Teaching is an isolating profession Most educators receive minimal feedback on their craft typically limited to one or two short visits from the administrator each year While some schools do encourage their teachers to observe peers the usual protocol is for new teachers to watch the veteran exemplary educators but those less experienced educators could benefit immensely from having peers observe them in action Teacher Driven Observation is unique in that it places the observed teacher as leader of the work the one who identifies a personal area for development engages her colleagues in the collection of classroom data and utilizes the data collected to inform her ongoing instruction Offers a new model for collecting data to inform and improve individual instruction Includes a step by step process for implementing Teacher Driven Observations Contains a comprehensive case study that clearly demonstrates the value of the Teacher Driven Observation model This book outlines and illustrates a step by step process for breaking down professional isolation and cultivating opportunities for meaningful learning through the Teacher Driven Observation process Te HS&T 2007 Shrt Crs M Holt Rinehart & Winston, 2007

Examining Lorraine Hansberry's A Raisin in the Sun as Counternarrative Carl A. Grant, 2023-09-29 Examining Lorraine Hansberry s A Raisin in the Sun as Counternarrative Understanding the Black Family and Black Students shows how and

why Lorraine Hansberry's play A Raisin in the Sun should be used as a teaching tool to help educators develop a more accurate and authentic understanding of the Black Family The purpose of this book is to help educators develop a greater awareness of Black children and youth s humanity academic potential and learning capacity and for teachers to develop the consciousness to disavow white supremacy American exceptionalism myths racial innocence and personal absolution within the education system This counternarrative responds to the flawed and racist perceptions stereotypes and tropes that are perpetuated in schools and society about the African American family and Black students in US schools It is deliberative and reverberating in addressing anti Black racism It argues that if Education is to be reimagined through a social justice structure teachers must be educated with works that include Black artists and educators and teachers must be committed to decolonizing their own minds Examining Lorraine Hansberry's A Raisin in the Sun as Counternarrative Understanding the Black Family and Black Students is important reading for undergraduate and postgraduate courses in Educational Foundations Curriculum and Instruction Education Policy Multicultural Education Social Justice Education and Black Studies It will also be beneficial reading for in service educators Intermediate Accounting Terry D. Warfield, Jerry J. Weygandt, Donald E. Kieso, 2007-12-04 Now readers can get all the accuracy and authority of the best selling intermediate accounting book in the new second edition of this brief streamlined version Fundamentals of Intermediate Accounting presents a balanced discussion of concepts and applications explaining the rationale behind business transactions before addressing the accounting and reporting for those activities Readers will gain a solid foundation in such areas as the standard setting process the three major financial statements revenue recognition income taxes reporting disclosure issues and much more **Resources in Vocational Education** ,1980 Facilitating Practitioner Research Susan Groundwater-Smith, 2012 This book addresses the complex dilemmas and issues that arise from the contribution made to practitioner inquiry Facilitating practitioner research is far more than providing advice upon methods which may be adopted or even modelling research practices and drawing attention to appropriate resources and theories It also requires the evolution of strong reciprocal partnerships that will contribute to professional knowledge formation in both the academy and the field When such engagement is undertaken then matters associated with authentic praxis development for field based and academic practitioners emerge The authors explore how a transformative partnership requires particular professional attitudes of practitioners and academic researchers and show how it should result in three integrated areas of knowledge the ideological the empirical and the technical The text draws upon extensive case studies from The Netherlands Sweden and Australia which not only to illustrate and illuminate but also highlight contradictions and tensions and serve to surface issues related to the quality of the partnerships between the academy and the field and the ways in which quality impacts upon practice These varying social geographies also allow a discussion of different intellectual traditions belief systems problem settings questions and discourses *Understanding Spoken English - Teacher's Book Two* Susan Boyer,2003 This book is

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