

12वीं जुन त्रैमासिक परीक्षा 2025

AFM/XII/10000

CLASS-XII MONTHLY EXAMINATION, JULY - 2nd

बाला - XII मासिक परीक्षा, जुलाई - 2025

CHEMISTRY (Elective)

रसायन शास्त्र (ऐच्छिक)

पृष्ठ : 1/4

(समय : 1 घंटा 30 मिनट)

1.56 Theory / वैयक्तिक

विषय कोड/Sub. Code :

Page : 1

(पूर्वक : 50)

[Time : 1 Hour 30 Minutes]

118

[Total Marks : 50]

Note : Read all the instructions carefully before attempting the questions.

Note : In case of any doubt or discrepancy in the questions, please refer to the original text.

Follow the instructions:

objective+ Subjective

Objective Type Questions

1. निम्नलिखित में से सही उत्तर चुनिए।/Choose the correct answer from the given options.

Question No. 1 to 30 have four options, out of which only one is correct. You have to mark your selected option, on the OMR Sheet with blue/black ballpoint pen only. Answer any 20 questions.

20 × 1 = 20

1. निम्नलिखित में से सबसे अधिक स्थिर आयन है।/The most stable ion in aqueous solution is

(A) V^{2+} (B) Ti^{2+} (C) Mn^{2+} (D) Cr^{2+}

2. लैंथेनाइड्स की सबसे प्रचलित ऑक्सीकरण अवस्थाएं हैं।/The most common oxidation states among the lanthanides are

(A) +1 (B) +2 (C) +4 (D) +3

3. समान आयन के जोड़े हैं।/The pairs of the same size are

(A) Zn^{2+} , Ni^{2+} (B) Fe^{2+} , Ni^{2+} (C) Zn^{2+} , Ti^{2+} (D) Zn^{2+} , W^{2+}

4. प्रत्येक लैंथेनाइड आयन में 14 f इलेक्ट्रॉन होते हैं।/The lanthanide ions have 14 f electrons.

12th Chemistry

Answer Key

23/06/2025

100% Correct

Answer Key Of Juet 2013 Of Chemistry

AW Rasmussen



Answer Key Of Juet 2013 Of Chemistry:

The book delves into Answer Key Of Juet 2013 Of Chemistry. Answer Key Of Juet 2013 Of Chemistry is a crucial topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Answer Key Of Juet 2013 Of Chemistry, encompassing both the fundamentals and more intricate discussions.

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Table of Contents Answer Key Of Juet 2013 Of Chemistry

1. Understanding the eBook Answer Key Of Juet 2013 Of Chemistry
 - The Rise of Digital Reading Answer Key Of Juet 2013 Of Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key Of Juet 2013 Of Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Key Of Juet 2013 Of Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key Of Juet 2013 Of Chemistry
 - Personalized Recommendations
 - Answer Key Of Juet 2013 Of Chemistry User Reviews and Ratings
 - Answer Key Of Juet 2013 Of Chemistry and Bestseller Lists
5. Accessing Answer Key Of Juet 2013 Of Chemistry Free and Paid eBooks
 - Answer Key Of Juet 2013 Of Chemistry Public Domain eBooks
 - Answer Key Of Juet 2013 Of Chemistry eBook Subscription Services
 - Answer Key Of Juet 2013 Of Chemistry Budget-Friendly Options
6. Navigating Answer Key Of Juet 2013 Of Chemistry eBook Formats
 - ePub, PDF, MOBI, and More
 - Answer Key Of Juet 2013 Of Chemistry Compatibility with Devices
 - Answer Key Of Juet 2013 Of Chemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer Key Of Juet 2013 Of Chemistry
 - Highlighting and Note-Taking Answer Key Of Juet 2013 Of Chemistry
 - Interactive Elements Answer Key Of Juet 2013 Of Chemistry

8. Staying Engaged with Answer Key Of Juet 2013 Of Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer Key Of Juet 2013 Of Chemistry
9. Balancing eBooks and Physical Books Answer Key Of Juet 2013 Of Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Key Of Juet 2013 Of Chemistry
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answer Key Of Juet 2013 Of Chemistry
 - Setting Reading Goals Answer Key Of Juet 2013 Of Chemistry
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answer Key Of Juet 2013 Of Chemistry
 - Fact-Checking eBook Content of Answer Key Of Juet 2013 Of Chemistry
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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