

Naming Acids and Bases Answers

Name the following acids and bases:

- | | | |
|----|--------------------------------|-----------------------------|
| 1) | NaOH | <u>sodium hydroxide</u> |
| 2) | H ₂ SO ₃ | <u>sulfurous acid</u> |
| 3) | H ₂ S | <u>hydrosulfuric acid</u> |
| 4) | H ₃ PO ₄ | <u>phosphoric acid</u> |
| 5) | NH ₃ | <u>ammonia</u> |
| 6) | HCN | <u>hydrocyanic acid</u> |
| 7) | Ca(OH) ₂ | <u>calcium hydroxide</u> |
| 8) | Fe(OH) ₃ | <u>iron (III) hydroxide</u> |
| 9) | H ₃ P | <u>hydrophosphoric acid</u> |

Write the formulas of the following acids and bases:

- | | | |
|-----|-----------------------|------------------------------------|
| 10) | hydrofluoric acid | <u>HF</u> |
| 11) | hydroselenic acid | <u>H₂Se</u> |
| 12) | carbonic acid | <u>H₂CO₃</u> |
| 13) | lithium hydroxide | <u>LiOH</u> |
| 14) | nitrous acid | <u>HNO₂</u> |
| 15) | cobalt (II) hydroxide | <u>Co(OH)₂</u> |
| 16) | sulfuric acid | <u>H₂SO₄</u> |
| 17) | beryllium hydroxide | <u>Be(OH)₂</u> |
| 18) | hydrobromic acid | <u>HBr</u> |

Answers For Naming Acids

Stephen R. Schaffter



Answers For Naming Acids:

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Chemistry Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the enhanced eText with online assessment answer specific responses and additional practice resources The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge Problems are arranged in an intuitive confidence building order

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