

8. Buffer, passing through the, is shown within the Earth's crust. Buffer is composed of the components of these of existing rocks.

12 samples of Buffer from a sequence were analysed to give the following composition of samples.

sample	composition (%)
1%	10.00
1%	10.00
1%	10.00

- (a) Define the term relative atomic mass.

The relative atomic mass of an element is the average mass of the atoms of the element, taking into account the relative abundance of the isotopes.

- (b) Calculate the relative atomic mass of the sample of buffer.

Give your answer to three decimal places.

$$\frac{(10 \times 10.00) + (10 \times 10.00) + (10 \times 10.00)}{30} = 10.00$$

$$\text{answer} = 10.00$$

- (c) Some studies, on early life, suggest that, in the early stages of life, the first molecules were made up of long particles which could be broken down into smaller molecules, which were of the same size as the particles of the early life forms.

Complete the table to show the number of subatomic particles in the ^{12}C atom and $^{12}\text{C}^{2+}$ ion.

	protons	neutrons	electrons
^{12}C	6	6	6
$^{12}\text{C}^{2+}$	6	6	4



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J Rink



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