

# Grade 11 Chemistry

## INTERMOLECULAR FORCES & Physical Properties

### EXAM QUESTIONS PART 3



| NAME OF SUBSTANCE | FORMULA                            | BOILING POINT (°C) |
|-------------------|------------------------------------|--------------------|
| Water             | H <sub>2</sub> O                   | 100                |
| Ethanol           | CH <sub>3</sub> CH <sub>2</sub> OH | 78                 |
| Bromine           | Br <sub>2</sub>                    | 58,8               |
| Iodine            | I <sub>2</sub>                     | 184,3              |
| Ammonia           | NH <sub>3</sub>                    | -33,3              |
| Phosphine         | PH <sub>3</sub>                    | -87,7              |

Explain why ethanol is soluble in water. Refer to the relative strength of the intermolecular forces in ethanol and water.

Explain why the boiling point of iodine is higher than that of bromine. Refer to the intermolecular forces present in EACH substance in the explanation.

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