

STOICHIOMETRY WORKSHEET

1. Calculate the number of moles of NaOH that are needed to react with 500 g of H_2SO_4 according to the following equation:



2. Calculate the mass of NH_3 that can be produced from the reaction of 125 g of NCl_3 according to the following equation:



3. Identify the limiting reactant and determine the mass of CO_2 that can be produced from the reaction of 25 g of C_3H_8 with 75 g of O_2 according to the following equation:

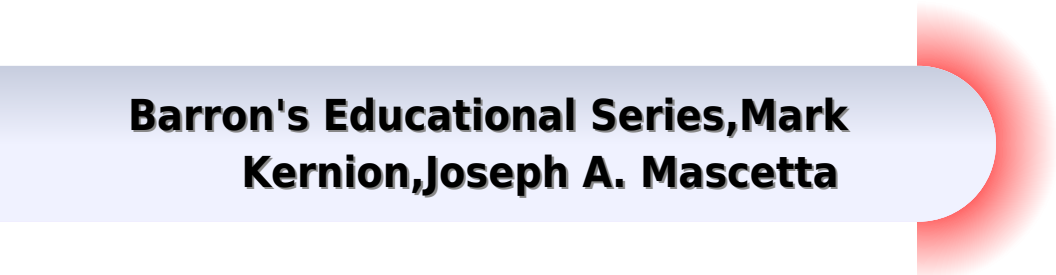


4. How many grams of SO_2 are produced when 152 g of CS_2 reacts with 48.0 g of O_2 according to the following equation:



Answers For The Chemistry Lab Of Stoichiometry

**Barron's Educational Series, Mark
Kernion, Joseph A. Mascetta**



Answers For The Chemistry Lab Of Stoichiometry:

Answers For The Chemistry Lab Of Stoichiometry Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the energy of words has be evident than ever. They have the capability to inspire, provoke, and ignite change. Such could be the essence of the book **Answers For The Chemistry Lab Of Stoichiometry**, a literary masterpiece that delves deep in to the significance of words and their impact on our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall effect on readers.

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