

Ternary compounds have _____ (three) _____ (three or) elements. They are usually always composed of a metal(s) _____ (metals) anionic ion(s) and a polyatomic _____ (polyatomic) anion. The only positively charged polyatomic ion is _____ (ammonium). To write the formula of a ternary compound is no different than to write the formula of a binary compound with one exception. If a subscript is necessary for the polyatomic ion in order to _____ (balance) charge, we must place the polyatomic ion in _____ (parentheses).

As each ion, write the formula of the ionic compound consisting of the cation on the left of the dash and the anionic polyatomic ion on the right.

	OH^-	SO_4^{2-}	NO_3^-	CO_3^{2-}	PO_4^{3-}
K^+	_____	_____	_____	_____	_____
Na^{2+}	_____	_____	_____	_____	_____
NH_4^+	_____	_____	_____	_____	_____
Al^{3+}	_____	_____	_____	_____	_____
Ca^{2+}	_____	_____	_____	_____	_____
Fe^{2+}	_____	_____	_____	_____	_____

Answers For Writing Ternary Formulas

**Tina Overton, Jonathan Rourke, Fraser
A. Armstrong**



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